Sequential fed-batch fermentation of 1,3-propanediol from glycerol by *Clostridium butyricum* DL07

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Abstract

The demand for 1,3-propanediol (1,3-PDO) has increased sharply due to its role as a monomer for the synthesis of polytrimethylene terephthalate (PTT). Although *Clostridium butyricum* is considered to be one of the most promising bio-producers for 1,3-PDO, its low productivity hinders its application on industrial scale because of the longer time needed for anaerobic cultivation. In this study, an excellent mutant of *C. butyricum* (DL07) was obtained with high-level titer and productivity of 1,3-PDO, i.e. 104.78 g/L and 3.38 g/(L*h) vs. 94.23 g/L and 3.04 g/(L*h) using pure or crude glycerol as substrate in fed-batch fermentation, respectively. Furthermore, a novel sequential fed-batch fermentation was investigated, in which the next bioreactor was inoculated by *C. butyricum* DL07 cells growing at exponential phase in the prior bioreactor. It could run steadily for at least eight cycles. The average concentration of 1,3-PDO in eight cycles was 84.62 g/L with the average productivity of 3.05 g/(L*h). The sequential fed-batch fermentation, and would greatly contribute to the industrial production of 1,3-PDO with higher productivity than repeated fed-batch fermentation, and would greatly contribute to the industrial production of 1,3-propanediol by *C. butyricum*. **Key words:** 1,3-Propanediol; glycerol; *Clostridium butyricum*; sequential fed-batch fermentation

INTRODUCTION

1,3-Propanediol is a versatile chemical compound with wide application in plastics, lubricants, cosmetics, pharmaceuticals and other fields. Especially, it is a crucial monomer for the synthesis of a novel biodegradable polymer, polytrimethylene terephthalate (PTT). PTT has been used in the textile industry for making carpets, apparel and fibre owing to its superior properties such as easy dye-ability, good resilience and softness(Kaur, Srivastava, & Chand,2012;Lee et al.,2015;Liu, Xu, Zheng, & Liu,2010). The demand of 1,3-PDO is growing rapidly with the profitable textile industry. At present, the production of 1,3-PDO can follow chemical and biological routes. Considering the mild reaction conditions and sustainable development of bioproduction, the bioconversion of 1,3-PDO has attracted great interest from researchers (Jiang, Wang, Wang, & Fang,2016;Saxena, Anand, Saran, & Isar,2009;Sun et al.,2018;Vivek, Pandey, & Binod,2017;Xiu & Zeng,2008).

Currently, glycerol, an alternative feedstock, is the most common substrate to produce 1,3-PDO for most 1,3-PDO-producing strains(Parate, Mane, Dharne, & Rode,2018). Interestingly, glycerol is an inevitable by-product of biodiesel, oleochemical industry, soap manufacturing and bioethanol production (Anitha, Kamarudin, & Kofli,2016;Rahim et al.,2020;Ringel, Wilkens, Hortig, Willke, & Vorlop,2012). Hopefully, for the valorization of glycerol, the surplus of waste glycerol could be reduced by the bioproduction of high value-added 1,3-PDO(Westbrook et al.,2019;Yang, Hanna, & Sun,2012). In nature, some specific microorganisms, including *Clostridium, Klebsiella*, *Citrobacter*, and *Enterobacter* (Chatzifragkou & Papanikolaou,2012), can consume glycerol to produce 1,3-PDO. Among them, *K. pneumoniae* and *C. butyricum* are the most

prominent 1,3-PDO producers considering their performance in high productivity and concentration of 1,3-PDO (Chatzifragkou & Papanikolaou,2012;Saxena et al.,2009). *C. butyricum* is thought to be the most potential strain to produce 1,3-PDO on an industrial scale because it is probiotics, anaerobic culture and vitamin B_{12} -independent (Cassir, Benamar, & La Scola,2016;Ringel et al.,2012;Yang et al.,2019). Thus many efforts have been made to accelerate the industrial production of 1,3-PDO by *C. butyricum*.

Many fermentation processes, such as batch, fed-batch, continuous, repeated batch and repeated fed-batch fermentation, were tried to produce 1,3-PDO(Guo et al.,2017;Oh, Lee, Heo, Seo, & Kim,2018a;Tee, Jahim, Tan, & Kim, 2017; Yang et al., 2017; Zhou et al., 2017; Zhou, Shen, Wang, Sun, & Xiu, 2018). Normally, fed-batch fermentation is recommended to be suitable for the industrial production due to its higher concentration of 1,3-PDO in all fermentation processes (Yang et al., 2018; Zhou et al., 2017). Further, repeated fed-batch fermentation was proposed to overcome some challenges in fed-batch fermentation, such as low overall productivity of 1,3-PDO and long time for seed culture (Oh, Lee, Heo, Seo, & Kim, 2018b:Xue, Li, Li, Xia, & Ye,2010). Repeated fed-batch fermentation is to pump out most of fermentation broths and remain an amount at the end of fed-batch fermentation, followed by fresh medium. To some extents, repeated fedbatch fermentation avoids the seed culture time between two fed-batches and enhances the overall products productivity, which is great promising for industrial 1,3-PDO production(Oh et al.,2018a). Considering the sustainable operations and overall productivity in industry, however, there are also some specific barriers and potential problems to repeated fed-batch fermentation. For example, the repeated fed-batch fermentations were investigated by K. pneumoniae only, which is pathogenic and vitamin B_{12} -dependent. Additionally, low 1,3-PDO concentration (66 g/L) (Xue et al.,2010) and more by-products were obtained (37.5 g/L 2.3propanediol) (Oh et al., 2018a). More importantly, the number of cycles is worrying, as the seed broths of each cycle come from the last cycle of fermentation, in which the microbial cells are at the decline phase. Undoubtedly, increasing the number of cycles and shortening the interval between the two cycles of inoculation will greatly reduce the infection risk of industrial production and further improve the overall 1.3-PDO productivity. Besides, reducing the production cost of the 1,3-PDO will make the factory profitable. The costs of carbon, organic nitrogen source and aerating nitrogen are the main obstacles to industrial production of 1,3-PDO(Ju et al.,2020;Wischral et al.,2016).

In this study, the production of 1,3-PDO by *C. butyricum* DL07 was evaluated in fed-batch fermentation using crude or refined glycerol as a substrate. Moreover, a sequential fed-batch fermentation process, whose seed broths was derived from the microbial exponential-phase of the last cycle fermentation, was proposed firstly to achieve semi-continuous and stable 1,3-PDO production by *C. butyricum*. It offers several advantages over previous fermentation processes, such as maintaining inoculum in good growth state, avoiding the individual culture of inoculation, shortening 1,3-PDO harvestable interval during every two cycles, and obtaining higher overall productivity. Certainly, an increase in number of cycles is expected to retain still stable 1,3-PDO concentration. Therefore, it is of great significant for industrial production to explore the production of 1,3-PDO in sequential fed-batch fermentation by *C. butyricum*. After that, the effects of corn steep liquor power and different aerating gases(N₂, H₂, CO₂, and fermentation exit gas) on the production of 1,3-PDO were studied so as to reduce the fermentation cost. Based on the optimization of fermentation conditions, it contributes to the 1,3-PDO production by microbial fermentation in industry. At the same time, this new fermentation process will facilitate further on a pilot and industrial scale.

MATERIALS AND METHODS

Material

Crude glycerol was provided by Sichuan Tianyu Oleochemical Co. Ltd., China. Its components include 78% glycerol, 0.87% ash, 15–17% moisture, and a little salt (the equivalent of electrical conductivity as 0.43% sodium chloride), and the pH value was 6.91 (Zhou et al.,2017). Active sludge was collected from an anaerobic digester at Dongtai Industrial Waste Treatment Co., Ltd., Dalian, China. The strain, *C. butyricum* DL07, was selected from the active sludge using seed medium and had been deposited in China General Microbiological Culture Collection Center (CGMCC, Beijing, China) and the accession number is CGMCC 17934. Bioreactors equipped with an automatic pH control system were requisite.

Culture media

The compositions of seed medium were similar to those described by Zhou et al.(Zhou et al.,2017). The solid seed medium for the selection of *C. butyricum* DL07 had the same chemicals as seed medium with addition of 1.2% agar. Fermentation medium for 1,3-PDO production was accordance with Jiang et al.(Jiang, Liu, Mu, Sun, & Xiu,2017), except that glycerol in the fermentation medium will be confirmed as either crude or refined glycerol according to the experimental needs. Moreover, the yeast extract in fermentation medium was replaced with corn steep liquor powder on the basis of experimental design. The feeding solution for 1,3-PDO production was glycerol or 80% glycerol with 40 g/L yeast extract.

Culture conditions

4% (v/v) of strain stored at -70°C was inoculated into 250 mL anaerobic serum bottles containing 100 mL of seed medium sterilized at 121°C for 15min and aerated by nitrogen. The seed culture was incubated at 37 °C and 200 rpm for 12 hours. Fermentations were carried out in a 5.0 L bioreactor (Baoxing Biotech, Shanghai, China) with 2.0 L sterilized fermentation medium. 10 % (v/v) of the seed culture broths was inoculated into the first bioreactor. N₂ was aerated into bioreactor to create an anaerobic environment. The bioreactor ran automatically at 37°C, 250 r/min, and pH 7.0 controlled by 5 mol/L NaOH solution throughout the fermentation process.

Selection and storage of strain

1 mL of anaerobic active sludge was mixed with 9 mL of sterile deoxidized water in serum bottle using a sterile syringe. The mixture was placed in a water bath at 80°C for 10 min to kill non-spore forming bacteria. The 100 μ L mixture was laid on the solid seed medium and cultured at 37°C in anaerobic incubator. After 12 h of culture, the largest colony of bacteria continued to be cultivated to purify the single bacteria. Based on serial purification, the single colony DL07 was finally isolated. Then the strain DL07 was enriched with liquid seed medium and cultured repeatedly using fermentation medium so as to improve its performance for 1,3-PDO production. The strain DL07 was identified as *C. butyricum*subsequently. Finally, *C. butyricum* DL07 was stored at -70°C in a seed medium with 40% glycerol.

Fed-batch fermentation of 1,3-PDO

Fed-batch fermentations were performed by means of two strategies: continuous and pulsed feeding. In continuous fed-batch fermentation, residual glycerol concentration was controlled at about 20 g/L during the fermentation by manually adding the feeding solution. In pulsed fed-batch fermentation, glycerol was added into bioreactor to reach the preset value for three times, when the residual concentration of glycerol was below 20 g/L.

The sequential fed-batch fermentation process was shown in Figure 1, in which the next bioreactor was inoculated by C. butyricum DL07 cells growing at exponential phase in the prior bioreactor. Firstly, the seed culture was scaled up three times. Then 2% of seed broth was added into the first bioreactor. When the first bioreactor ran to 10 h, 40 mL fermentation broth (2% inoculation) was inoculated to the second bioreactor to start the second cycle of fermentation. Meanwhile, the first bioreactor proceeded as a fed-batch fermentation. The following fermentations were carried out in the same way. In order to study aerating gas fermentation, the exit gas produced in the first bioreactor was aerated directly into the second bioreactor before inoculation to create an anaerobic environment.

Analytical methods

The cell mass was monitored by measuring the optical density of the cultured broths at 650 nm with appropriate dilution using a UV-visible spectroscopy system. Glycerol was determined by adding specific NaOH solution into a reaction solution where glycerol reacted with excess sodium periodate and then unreacted sodium periodate was reacted with ethylene glycol (Wang et al. 2001). The concentration of glycerol, 1,3-PDO, butyrate, acetate, lactate, and formate were determined by HPLC analysis system (Waters 1515) using an Aminex HPX 87H column (300 mm \times 7.8 mm; Bio-Rad) coupled with an autosampler (Waters 2707)

and a differential refractometer (Waters 2414). Operating conditions: sample volume 20 μ L, mobile phase 5 mmol/L H₂SO₄, flow rate 0.6 mL/min, detector temperatures 35°C, column temperature 65°C. Testing samples, which are obtained from the fermentation broths after centrifuging for 10 minutes at 12,000 r/min, were diluted appropriately and filtered through 0.22- μ m membranes filter before the test.

Statistical analysis

Linear regression was accomplished using Microsoft Excel software after collecting a large amount of data point. A significance test was recommended using two-tailed paired Student's t-test, and p <0.05 was considered significant.

RESULTS

Fed-batch fermentation of a novel strain

To evaluate the ability of C. butyricum DL07 to produce 1,3-PDO, fed-batch fermentation was performed with two feeding strategies. Meanwhile, pure and crude glycerol were investigated as feeding solution respectively. Fed-batch fermentation coupled with continuous feeding was conducted with an initial glycerol concentration of 40 g/L and then 20 g/L during the fermentation. Nitrogen was pumped into bioreactor at 0.15 vvm for 1 h before and after inoculation to create an anaerobic environment. As shown in Figure 2, the rate of glycerol consumption and 1,3-PDO production rose rapidly between 3-10 h, so the highest 1,3-PDO productivity (10.43 g/(L[?]h)) occurred in this period. The final 1,3-PDO concentration was up to 85.96 g/L with pure glycerol and 85.47 g/L with crude glycerol. Respectively, the overall 1,3-PDO productivity reached 2.87 and 2.85 g/(L[?]h) with the yield of 0.520 $g_{1,3-PDO}/g_{Gly}$. Whether crude or refined glycerol as substrate, the highest OD value, representing the maximum biomass, appeared at 13 h with OD value of 15.76 and 14.73, respectively. Additionally, butyric acid was the most by-product with concentrations ranging from about 15.11 to 16.65 g/L. The concentrations of acetic acid, followed by butyric acid, were about 9.22-9.71 g/L. When glycerol was pumped by three-pulse strategy, the concentrations of 1,3-PDO and organic acid approached that in continuous fed-batch fermentation. And 84.98 and 84.86 g/L 1,3-PDO were produced using pure and crude glycerol as substrate, respectively. Similarly, the 1,3-PDO productivity was 2.83 g/(L[?]h) corresponding with the yield of 0.521 $g_{1.3-PDO}/g_{Glv}$ either pure or crude glycerol. These values were also at the same level with those of fed-batch fermentation by continuous feeding. The highest OD value also appeared at 13 h but decreased slightly compared to continuous fed-batch fermentation.

Effect of organic nitrogen resource on the production of 1,3-PDO

The costs of nitrogen and carbon source are the key factors in 1,3-PDO production. In fed-batch fermentation, 1,3-PDO concentration by pure glycerol fermentation was similar with that by crude glycerol fermentation. As a result, refined glycerol can be substituted by crude glycerol for 1,3-PDO production so as to reduce the cost of 1,3-PDO production. Furthermore, the addition of yeast extract and core steep liquor powder was investigated. When the feeding solution consisted of 80% glycerol and 40 g/L yeast extract, continuous fed-batch fermentation was performed with 0.1 vvm nitrogen through fermentation. Unexpectedly, the concentration of 1,3-PDO increased significantly whether crude or refined glycerol as substrate. Using pure glycerol and yeast extract as feeding solution, the maximum 1,3-PDO concentration was in surprise up to 104.78 g/L (Figure 3a) with a productivity of 3.38 g/(L[?]h), which was 21.9% higher than that using only glycerol as the feeding solution. The by-products and biomass also had relative high concentrations. Specifically, 19.63 g/L butyrate and 10.52 g/L acetate were obtained at the end of fermentation. The 1,3-PDO conversion rate was $0.539 \text{ g}_{1.3-\text{PDO}}/\text{g}_{\text{Gly}}$. However, when using crude glycerol and yeast extract as feeding solution, the concentration of 1,3-PDO and butyrate decreased to 94.23 and 14.87 g/L (Figure 3b) respectively. Acetate concentration (12.90 g/L) was higher. The highest OD value was 23.3 % lower compared with that of pure glycerol fermentation. Lastly, an overall productivity was 3.04 g/(L*h) with a conversion of 0.522 $g_{1,3-PDO}/g_{Gly}$.

Corn steep liquor powder, an inexpensive nitrogen source, can be used as organic nitrogen instead of yeast extract for 1,3-PDO production on account of vitamin B_{12} -independent in *C. butyricum*. Fed-batch fermen-

tations by three-pulse strategy with crude glycerol were conducted using 4 and 6 g/L corn steep liquor power as organic nitrogen source. The results were shown in Figure 4. When the concentration of corn steep liquor power was 4 g/L, respectively, the concentrations of 1,3-PDO, butyrate and acetate were 82.05, 15.03 and 11.32 g/L with no obvious difference in contrast to yeast extract as nitrogen source in fermentation. And there was no significant fluctuation in 1,3-PDO productivity and yield with the change of nitrogen sources. Otherwise, microbial growth was slightly slower. When 6 g/L corn steep liquor power was added in medium, the concentration of 1,3-PDO went up to 88.25 g/L with the increase of biomass. It seems corn steep liquor powder as a perfect nitrogen substitute can significantly reduce the production cost of 1,3-PDO to promote the industrial production of 1,3-PDO.

Effect of aerating gas on the production of 1,3-PDO

Nitrogen is used extensively in the conversion of glycerol to 1,3-PDO by C. butyricum, since C. butyricum is anaerobic bacteria and undergoes anaerobic fermentation. It is all known that CO_2 and H_2 are produced by C. butyricum during the glycerol fermentation. The produced CO_2 is of no value and its emissions to environment can put pressure on environmental protection. Hence, if the gas produced by fermentation could be substituted for nitrogen for the next fermentation, it would avoid the cost of carbon dioxide treatment and reduce the use of nitrogen. Fed-batch fermentations were conducted, in which gas produced by fermentation was used to create an anerobic environment instead of nitrogen. The gas from a traditional fed-batch fermentation before 10 h passed through the second fermentation. Then the second fermentation started with 10% inoculation. As shown in Figure 5, the concentration of 1,3-PDO (76.58 g/L) in the second fermentation (fermentation exit gas as an aeration gas) was slightly lower than that (82.60 g/L) in the first fermentation (N₂ as an aeration gas), but the productivity of 1,3-PDO (3.38 g/(L[?]h)) increased by 23%. Unfortunately, the conversion rate of 1,3-PDO was only 0.481 $g_{1,3-PDO}/g_{Glv}$, which was lower than that of the first fermentation $(0.513 g_{1,3-PDO}/g_{Gly})$. On the contrary, the concentration of butyric acid was higher. Then, H_2 and CO_2 took the place of N_2 to explore their effects on fed-batch fermentation, respectively. The results (Figure 5) showed that 1,3-PDO, butyrate and acetate concentrations (81.34, 15.24 and 10.50 g/L) remained a same level whether hydrogen or nitrogen for creation of anaerobic fermentation environment. Similarly, the productivity and conversion rate of 1,3-PDO (2.80 g/(L[?]h) and $0.520 \text{ g}_{1,3-\text{PDO}/\text{g}}$ _{Gly}) remained basically unchanged. However, when CO_2 was used to create an erobic environment, the concentration of 1,3-PDO (78.77 g/L) showed a slight decrease compared to N₂ pumped into bioreactor, but was essentially similar with fermentation exit gas as aeration gas. And other by-products concentrations (15.95 g/L butyrate and 8.52 g/L acetate), 1,3-PDO conversion rate $(0.477 \text{ g}_{1.3-\text{PDO}}/\text{g}_{\text{Glv}})$ and productivity were consistent with those of fermentation exit gas as aeration gas.

sequential fed-batch fermentation

The sequential fed-batch fermentation was carried out followed the Figure 1 with three-pulse feeding strategy. To meet the demand of large quantities of seeds in industrial production, seed culture was scaled up three times in three bioreactors. In seed culture process, the inoculation was carried out in sequence, when OD values were 3.020, 3.060 and 5.456 in three bioreactor, respectively, with the corresponding seed culture time of 4, 4.5 and 7 h. As a consequence, 15.5 h was needed to scale up the seed culture before fermentation. There is no doubt that seed culture takes some time in industrial production, and a lot of money also needs to be invested, which will reduce the benefits.

However, after seed culture, if sequential fed-batch fermentation was initiated, then no further seed culture was required. Four bioreactors were required to produce 1,3-PDO with a seed culture of 10 h simultaneously by the sequential fed-batch fermentation , as it takes about 30 h to finish fermentation in a fed-batch fermentation. When the first fermentation ran until 10 h, only 2% of the fermentation broths was inoculated into a new bioreactor with fresh fermentation medium. Then, the process continued in cycles. As a result, the fermentation ran steadily for at least eight cycles and was stopped by human intervention due to lack of technicians to feed glycerol. The concentrations of 1,3-PDO with corresponding productivities, yields even by-products concentrations from the first to eighth cycles were listed in table 1. The results showed that the highest concentration of 1,3-PDO (88.59 g/L) occurred in the second cycle of fermentation. While the

lowest 1,3-PDO concentration (80.65 g/L) was obtained in the eighth cycle of fermentation. This fluctuation in concentrations of 1,3-PDO is acceptable by factory because more than 80 g/L 1,3-PDO is worthy for industrial production. It can be seen from calculation results that the highest and lowest productivities were 3.83 and 2.74 g/(L[?]h) in the fifth and third cycle, respectively. The minimum productivity in sequential fed-batch fermentation remained at a same level with the single fed-batch fermentation. The productivities of the second and fifth fermentation were 3.52 and 3.83 g/(L[?]h), respectively, which was much higher than the result of single fed-batch fermentation. Overall, the average concentrations of 1,3-PDO, butyrate and acetate were 84.62, 16.76 and 7.28 g/L. Beyond that, the average productivity was 3.05 g/(L[?]h) with a conversion of 0.52 g_{1,3-PDO}/g_{Gly}. It is not negligible that the residual glycerol concentration was controlled below 10 g/L in every fermentation. It's worth noting that the sequential fed-batch fermentation only took 100 h to finish eight fed-batch fermentations without additional seed culture.

Glycerol feeding strategy based on NaOH consumption in fed-batch fermentation

In fed-batch fermentation, glycerol was added into bioreactor manually. Obviously, an automatic feeding system will facilitate the application of 1,3-PDO production in industry. In order to develop glycerol feeding automatically, a large amount of data including the consumption of 5 mol/L NaOH and glycerol, the production of 1,3-PDO, and bacterial growth were collected in sequential fed-batch fermentation to explore their relationships. Fortunately, the specific mass consumption of glycerol showed a good linear relationship with 5 mol/L NaOH (R^2 =0.976, p=0.006) (Figure 6). 5 mol/L NaOH was automatically pumped into the bioreactor coupling with a pH control system, so the mass consumption of 5 mol/L NaOH could be calculated on the basis of the times of delivery recorded in the bioreactor. It was calculated that a total of 198.5-208.5 g/L NaOH solution (5 mol/L) was pumped into the bioreactor to maintain the pH (7) during a fed-batch fermentation. Based on their linear relationship, the mass of glycerol feeding could be calculated according to the consumption of 5 mol/L NaOH. In this way, glycerol feeding could proceed automatically during fermentation.

To practice the feasibility of this linear relationship, fed-batch fermentations were carried out in which glycerol was fed into the bioreactor in terms of the consumption of 5 mol/L NaOH to conform their linear relationship. When the initial glycerol concentration declined from 40 to 20 g/L, glycerol feeding system was activated in accordance with preset procedure. As predicted, the concentration of glycerol was maintained at about 20 g/L with a small fluctuation (Figure 7), which did not interfere with normal fermentation. Because glycerol feeding was ceased at 20 h of fermentation, the residual glycerol concentration at the end of fermentation was 5.97 g/L. The concentration of 1,3-PDO was 82.18 g/L corresponding with a productivity of 2.74 g/(L[?]h). In addition, the butyrate, acetate and cell concentrations were at a same level compared with those of manual feeding, so did the 1,3-PDO yield. In a word, the automatic glycerol feeding system could apply to the production of 1,3-PDO by *C. butyricum* DL07 fermentation.

1. DISCUSSION

2. High-level production of 1,3-PDO by C. butyricumDL07

Fed-batch fermentation, which is an effective fermentation process to achieve high product concentration, gets extensive attention in microbial fermentation and is feasible for industrial production(Lee et al.,2015). However, the concentration and yield of 1,3-PDO are subject to fluctuation with different feeding methods in fed-batch fermentation(Yang et al.,2018). Thus, two glycerol feeding methods (continuous feeding and three-pulse feeding) were adopted to perform the production of 1,3-PDO by *C. butyricum* DL07. At the same time, the 1,3-PDO production was assessed with pure and crude glycerol as substrate using two feeding methods, respectively. When feeding solution only contained glycerol in fed-batch fermentation (85.96 g/L 1,3-PDO) or crude glycerol (85.47 g/L) as substrate. The same results were validated, although three-pulse feeding was adopted in fed-batch fermentation. In addition, the concentrations of by-products, 1,3-PDO productivity and yield did not appeal to be affected by the quality of glycerol. These suggested that the presence of few impurities in crude glycerol did not affect the production of 1,3-PDO by *C. butyricum* DL07. Similarly, the concentration of 1,3-PDO and other by-products did not fluctuate significantly with

the two different feeding methods, whether crude or refined glycerol as substrate. These results indicated that *C.butyricum* DL07 possess a certain robustness and can produce 1,3-PDO steadily and efficiently under different fermentation conditions. Overall, the concentrations of 1,3-PDO were among 82.76-88.17 g/L with a subtle difference in productivities (2.74-2.85 g/(L[?]h)) in fed-batch fermentation by C.*butyricum* DL07. And the convention rates of 1,3-PDO ranged from 0.517 to 0.523 g_{1,3-PDO}/g_{Gly}. It can be seen that this superior *C. butyricum* is suitable for industrial production of 1,3-PDO because of its excellent production performance.

When the mixture of yeast and glycerol was used as feeding solution, 93.7 g/L 1.3-PDO was produced by natural C. butyricum in previous report(Wilkens, Ringel, Hortig, Wilke, & Vorlop, 2012). In order to motivate the maximum potential of C. butyricum DL07 to produce 1,3-PDO, the same feeding solution without concern for cost was used for the production of 1,3-PDO in fed-batch fermentation. Surprisingly, 104.78 and 94.23 g/L 1.3-PDO were obtained with pure and crude glycerol as substrate with an overall productivity of 3.38 and 3.04 g/(L*h), respectively. To our knowledge, these are the highest 1,3-PDO concentration for natural producers so far. Some examples for 1,3-PDO production by natural producers are given in Table 2. The maximum tolerance concentration of C. butyricum to 1,3-PDO was reported to be 83.7 g/L (Colin, Bories, & Moulin, 2000). It can be assumed that the presence of amino acids and vitamins in yeast extract has positive effects on strain for tolerance of the high concentration of 1,3-PDO. It was reported that high-level amino acids in cell were used to response to the deterioration of living environment (Zhang, Liu, Huang, Fu, & Fang, 2018). Moreover, the increased yeast extract in fermentation promoted microbial growth. It has been reported that the addition of amino acids could improve cells growth (Sharma & Melkania, 2018). Obviously, the concentration of 1,3-PDO was relatively lower using crude glycerol, possibly due to the dilution and the accumulation of impurities with the addition of a large amount of crude glycerol to the bioreactor. After all, a microorganism has a certain tolerance for impurities in crude glycerol. It has been reported a certain concentration of salts, fatty acids, heavy metal ions have inhibitory effects on C. butyricum (Chatzifragkou, Dietz, Komaitis, Zeng, & Papanikolaou, 2010; Matsumura, Nomura, & Sato, 2008).

Reduction in 1,3-PDO production cost

The high production cost of 1,3-PDO is one of the factors hindering its industrial production(Zeng & Sabra,2011). In order to reduce production costs, the residual glycerol from biodiesel production was used to replace pure glycerol for 1,3-PDO production (Chatzifragkou et al.,2014;Xiu, Wang, Zhou, & Sun,2019). Besides, soybean cake hydrolysate and corn steep liquor as organic nitrogen in medium were tried to substitute the expensive yeast extract (Maina et al.,2019;Oh et al.,2018a;Wischral et al.,2016). We have concluded that crude glycerol can be used as a substrate for fermentation because of the robustness of *C. butyricum* DL07. Therefore, corn steep liquor powder, as a primary by-product of corn-starch extraction process, was expected to support the growth and fermentation of *C. butyricum* DL07. The results showed that corn steep liquor powder can be used as organic nitrogen source without the loss of 1,3-PDO concentration, yield and productivity. By calculating the cost of production, it is more economical to add 4 g/L corn steep liquor powder in fermentation medium. The concentration of corn steep liquor power is relatively lower compared with an addition of 10% in a report (Oh et al.,2018a) and 30 g/L in another report (Wischral et al.,2016).

The fermentation exist gas including more than 95% CO₂ and a little H₂ was produced according to the glycerol metabolism pattern of *C. butyricum* for the 1,3-PDO production (Saint-Amans, Girbal, Andrade, Ahrens, & Soucaille,2001;Zhang et al.,2018). Furthermore, the fermentation exist gas instead of nitrogen creating an anaerobic fermentation environment was performed. Unfortunately, the concentration and conversion rate of 1,3-PDO were lower. In contrast, the productivity of 1,3-PDO and butyric acid concentration were higher. H₂ and CO₂were then used to substitute nitrogen for fed-batch fermentation to explore which gas affected the results. It turned out that the existence of CO₂, not H₂, affected the production of 1,3-PDO. The presence of CO₂ reduced the concentration. This indicated that CO₂ causes glycerol metabolism to flow to butyric acid production. It has been reported that CO₂ can promote the growth of *C. butyricum*, the growth will be delayed or does not start at all if without CO₂(Hakalehto & Hänninen,2012). In the future,

an appropriate mixture of fermentation exist gas and nitrogen used in fermentation is expected to increase productivity without losses of 1,3-PDO concentration and conversion rate. At the same time, there will be a little reduction in the production cost of 1,3-PDO, and the pressure of fermentation exhaust emissions on the environment will also be relieved to a certain extent.

Stable and efficient sequential fed-batch fermentation

From the perspective of microbial growth (Figure 2), the exponential growth period of the strain was between 3 h and 13 h in all fed-batch fermentations. As is well known, fermentation will be performed more steadily and successfully, when microbes are used as inoculum at exponential growth period rather than decline phase. In repeated batch and fed-batch fermentations, a majority of fermentation broths were removed, leaving a part of microbes during decline phase as seed in the bioreactor, then a certain fresh fermentation medium was added into the bioreactor (Kaur, Srivastava, & Chand,2012;Oh et al.,2018a;Szymanowska-Powalowska,2014;Yang et al.,2017). It has been reported that the performance of strain to produce 1,3-PDO decreased as cycle increased (Chatzifragkou et al.,2018). Therefore, the times of cycle is worrying, especially for industrial production although five cycles were carried out successfully with seeds in the decline phase (Oh et al.,2018a). Furthermore, there is no report about repeated fed-batch fermentation by C. butyricum. It is necessary to alleviate the anxiety of repeated fed-batch fermentation with seeds in the decline phase.

A sequential fed-batch fermentation, in which C. butyricum DL07 in exponential growth period was inoculated into a new fermentation in turn, was proposed in this study (Figure 1). As predicted, the experiments were carried out for eight cycles successfully without a significant decrease in concentration of 1,3-PDO. The concentrations of by-products (butyrate and acetate) also did not appear significant fluctuations. The trends of 1,3-PDO production and microbial growth were similar in eight separate fermentations. Especially during the first 10 h of production, it can be seen that the formation of 1.3-PDO was stable. For this reason, the strain cultured in bioreactor for 10 h was reliable to be inoculums. It is worth mentioning that only 2% fermentation broth was inoculated into the next cycle in sequential fed-batch fermentation. The inoculation was generally 5%-10% in most of the 1,3-PDO fermentations (Chatzifragkou et al.,2011,2014;Lee, Jung, & Oh,2018; Yang et al., 2018). Compared with traditional fed-batch fermentation, the stable 1,3-PDO concentration was obtained in repeated fed-batch fermentation with 10% inoculation(Oh et al., 2018a:Xue et al., 2010). There is no doubt that 2% inoculation without negative effects on 1,3-PDO production will bring more benefits to enterprises. On the whole, in sequential fed-batch fermentation, the average concentration of 1,3-PDO in eight cycles was 84.62 g/L with the average productivity of 3.05 g/(L*h). In this process, the time of seed cultivation in traditional fermentation was avoided, and the potential problem of fewer cycle times caused by poor seed quality in repeated batch and fed-batch fermentation was solved as well. Successfully, semi-continuous production of 1.3-PDO was achieved in this process. In a word, the sequential fed-batch fermentation is a stable process for 1.3-PDO production so as to be used for industrialization.

The production of 1,3-PDO with an automatic glycerol feeding system

Regrettably, the glycerol was manually pumped into the bioreactor in the sequential fed-batch fermentation. Generally, measuring glycerol concentration in time is required for two feeding strategies (continuous feeding and pulse feeding) in fed-batch fermentation. In continuous feeding, glycerol is manually fed into the bioreactor with a constant or varying rate to remain a certain range of concentration(Chen et al.,2003;Jiang et al.,2017). And in pulse feeding, whenever the glycerol concentration decreases to a preset value, the glycerol is manually added to the bioreactor to reach a specific concentration(Chatzifragkou et al.,2014;Hiremath, Kannabiran, & Rangaswamy,2011;Zhou et al.,2017). This is undoubtedly a technical labor burden for industrial production and increases the instability of fermentation. There are some automatically feeding strategies for 1,3-propanediol fermentation by K. pneumoniae . For example, glycerol feeding is coupled with the consumption of alkali and fermentation time(Huang, Chen, Sun, & Liu,2015). And coupling glycerol feeding with the alkali consumption and the production of carbon dioxide has been studied (Song, TENG, & Xiu,2012). In another report, a method was developed of coupling glycerol feeding and ammonium with alkali consumption (Reimann & Biebl,1996). These methods were applied based on two relations. However,

these relationships do not apply to C. butyricum DL07, so a simple and easy-to-operate automatic feeding strategy is needed.

In sequential fed-batch fermentation, a large amount of data was collected and used to explore their relationship. Finally, a good linear relationship was established between the mass consumption of glycerol and 5M NaOH solution (Figure 6). The pumping times of 5M NaOH were recorded by the bioreactor system. So the automatic glycerol feeding can be realized by coupling a computer with a bioreactor system. Of course, this linear relationship must be put into the computer. In this way, glycerol can remain a certain concentration during fermentation by the automatic feeding system. When the glycerol fed in bioreactor reaches a set value, the feed will be stopped so as to remain the glycerol concentration within a reasonable range at the end of fermentation. This set value is calculated based on the final 1,3-PDO concentration and conversion rate in fed-batch fermentation. After all, the residual glycerol concentration should be low enough to relieve the pressure of separating 1.3-PDO from glycerol (Xiu et al., 2004). To further verify the applicability of the linear relationship we obtained, fed-batch fermentation was subsequently performed. Obviously, when the initial glycerol concentration decreased to 20 g/L, the concentration of glycerol remained 18.44-21.66 g/L by automatic glycerol feeding strategy. As predicted, when the fermentation was finished, the residual glycerol concentration in bioreactor was relatively lower (5.97 g/L) and 1,3-PDO concentration, conversion rate and productivity maintained at the same level compared with manual glycerol feeding. This means that the automatic feeding system is well-suited for the fermentation of C. butyricum DL07. This automatic feeding system provides an operator-independent, repeatable and highly desirable process for 1,3-PDO production by C. butyricum DL07 fermentation from lab-scale to industrial scale. In the future, the novel sequential fed-batch fermentation with an automatic glycerol feeding system will have an opportunity to be used in industrial production of 1,3-PDO.

CONCLUSION

C. butyricum DL07 possess a certain robustness, and it can convert pure and crude glycerol to 1.3-PDO by using two feeding strategies in fed-batch fermentation. And the concentration of 1,3-PDO was stable and higher than 80 g/L. When the feeding solution was a mixture of yeast and glycerol, to our surprise, 104.78 and 94.23 g/L 1,3-PDO were obtained using pure and crude glycerol as substrate with an overall productivity of 3.38 and 3.04 g/(L*h), respectively. To our knowledge, these are among the best results published so far for natural producers. Then, corn steep liquor powder as an inexpensive alternative nitrogen source can be used for 1,3-PDO production to reduce the production cost. In addition, it is expected that fermentation exist gas and nitrogen are used as aerating gas in a suitable proportion so as to increase 1,3-PDO productivity without the loss of concentration and conversion rate. In this way, the cost of fermentation and pressure on environment caused by the exhaust will be reduced at the same time. More importantly, a sequential fed-batch fermentation process was established to achieve a stable, semi-continuous 1,3-PDO production. In this process, the average concentration of 1,3-PDO, 84.62 g/L, was obtained with the average productivity of 3.05 g/(L*h). This process avoids seed cultivation and solves the potential problem of fewer cycle times caused by poor seed quality in repeated fed-batch fermentation. Furthermore, an automatic glycerol feeding strategy was applied in the production of 1.3-PDO by coupling the mass consumption of glycerol with 5M NaOH solution. Therefore, the novel sequential fed-batch fermentation with an automatic glycerol feeding system will have an opportunity to be used in the industrial production of 1.3-PDO. This sequential fed-batch fermentation with an automatic glycerol feeding system and optimization of fermentation may be valuable for reducing the cost of 1,3-PDO production and promote the industrial production of 1,3-propanediol by C. butyricum.

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TABLE 1 Sequential fed-batch fermentations by C. butyricum DL07

Sequential fermentations	1,3-PDO (g/L)	Butyrate (g/L)	Acetate (g/L)	Lactate (g/L)	$\begin{array}{c} Q_{1,3\text{-PDO}} \ (g \ L^{-1} \ h^{-1}) \end{array}$	Conversion rate (g $_{1,3-PDO}$ /g $_{Gly}$)
Seed culture I	6.30	0.99	0.72	1.01	-	-
Seed culture II	6.98	0.74	1.15	0.98	-	-
Seed culture III	14.69	1.88	2.08	1.46	-	-
Fermentation I	84.04	17.53	6.59	2.91	2.80	0.52
Fermentation II	88.59	17.50	7.80	3.59	3.54	0.54
Fermentation III	82.20	16.39	7.76	4.12	2.74	0.50
Fermentation IV	86.30	17.31	3.98	3.02	2.88	0.53
Fermentation V	88.18	17.24	8.29	4.32	3.83	0.54
Fermentation VI	82.31	16.35	7.79	4.18	2.84	0.50
Fermentation VII	84.65	16.53	8.03	2.54	3.02	0.52
Fermentation VIII	80.65	15.22	7.96	2.19	2.78	0.49
Average	$84.62 {\pm} 2.89$	$16.76 {\pm} 0.79$	7.28 ± 1.42	$3.36{\pm}0.81$	$3.05 {\pm} 0.41$	$0.52{\pm}0.02$

TABLE 2 Summary of high-level 1,3-PDO production from natural producers in fed-batch fermentations

						Overall produc- tivity				
Microorga	Titer	Yield $(g. g^{-1})$	Yield (g. g^{-1})	Yield (g. g^{-1})	Yield (g. g^{-1})	(g.L ⁻¹ .h ⁻¹)	Glycerol source	Glycerol source	Glycerol source	Ref.
C. bu- tyricum IK 124	87.7	87.7	0.54	0.54	1.90	1.90	1.90	Refined	(Hirschma Baganz, Koschik, &	an(Hirschma Baganz, Koschik, & 05¥orlop,200
	80.1	80.1	0.56	0.56	1.80	1.80	1.80	Crude	1,	, 1,
C. bu- tyricum AKR 102a	93.7	93.7	0.52	0.52	3.30	3.30	3.30	Refined	(Wilkens et al.,2012)	(Wilkens et al.,2012)
	76.2	76.2	0.51	0.51	2.30	2.30	2.30	Crude		
C. bu- tyricum DSM 5431	70.3	70.3	0.68	0.68	1.50	1.50	1.50	Crude	(Matsumu et al.,2008)	ır ∜ Matsumu et al.,2008)

Microorgar	Titer n(sgnL ⁻¹)	Yield (g. g^{-1})	Yield (g. g^{-1})	Yield (g. g ⁻¹)	$\begin{array}{l} \text{Yield} \\ (\text{g. } \text{g}^{-1}) \end{array}$	Overall produc- tivity (g.L - ¹ .h ⁻¹)	Glycerol source	Glycerol source	Glycerol source	Ref.
C. bu- tyricum VPI 1718	67.9	67.9	0.55	0.55	0.78	0.78	0.78	Crude	(Chatzifra et al.,2011)	g k6h atzifra et al.,2011)
K. pneu- mo- niae DSM 4799	80.2	80.2	80.2	0.45	1.16	1.16	1.16	Crude	(Jun et al.,2010)	(Jun et al.,2010)
K. pneu- mo- niae LX3	71.38	71.38	71.38	0.49	2.24	2.24	2.24	Refined	(Xue et al.,2010)	(Xue et al.,2010)
K. pneu- mo- niae ME- 303	67.21	67.21	67.21	0.52	1.82	1.82	1.82	Refined	(Jin, Lu, Huang, Luo, & Li,2011)	(Jin, Lu, Huang, Luo, & Li,2011)
K. pneu- mo- niae ME- 308	70	70	70	0.58	0.97	0.97	0.97	Refined	(Ji, Huang, Zhu, Hu, & Li,2009)	(Ji, Huang, Zhu, Hu, & Li,2009)
K. pneu- mo- niae DSM2026	61.9	61.9	61.9	0.4	2.0	2.0	2.0	Refined	(Mu, Teng, Zhang, Wang, & Xiu,2006)	(Mu, Teng, Zhang, Wang, & Xiu,2006)
C. fre- undii FMCC- B294	68.1	68.1	68.1	0.40	0.79	0.79	0.79	Crude	(Metsoviti Zeng, Kouti- nas, &	, (Metsoviti Zeng, Kouti- nas, & a Raj2013 }ol:
L. di- olivo- rans DSM 14421	85.4	85.4	85.4	0.46	0.46	0.46	0.46	Refined	(Pflügl, Marx, Mat- tano- vich, &	(Pflügl, Marx, Mat- tano- vich, & 4)Sauer,2014

Microorgai	Titer n(ggnL ⁻¹)	Yield (g. g ⁻¹)	Yield (g. g ⁻¹)	Yield (g. g ⁻¹)	Yield (g. g^{-1})	Overall produc- tivity (g.L - ¹ .h ⁻¹)	Glycerol source	Glycerol source	Glycerol source	Ref.
L. reuteri ATCC 55730	65.3	65.3	65.3	0.16	1.20	1.20	1.20	Refined	(Jolly, Hitz- mann, Ra- ma- lingam, & Ramachar	(Jolly, Hitz- mann, Ra- ma- lingam, & mcRam;20ch4)
butyricum DL07	104.8 94.2	104.8 94.2	104.8 94.2	0.54 0.52	3.38 3.04	3.38 3.04	3.38 3.04	Refined Crude	This study	This study

Figures Legends

FIGURE 1 The schematic process of sequential fed-batch fermentation

FIGURE 2 Fed-batch fermentation of 1,3-propanediol by *C. butyricum* DL07 using pure glycerol (A, C) and crude glycerol (C, D). Two feeding strategies were adopted including continuous feeding (A, B) and pulsed feeding (C, D). Values are means of two independent fermentations

FIGURE 3 Fed-batch fermentation by *C. butyricum* DL07 feeding a mixture of yeast and pure glycerol (A) or crude glycerol (B)

FIGURE 4 Fed-batch fermentation using corn steep liquor powder instead of yeast extract as organic nitrogen resource

FIGURE 5 Fed-batch fermentation by *C. butyricum* DL07 with different aerating gas supplies for 1 h before and after inoculation, respectively

FIGURE 6 The relationship between consumption of 5M NaOH and glycerol

FIGURE 7 Fed-batch fermentation feeding automatically pure glycerol according to the linear relationship obtained in Figure 6





