

Tracing the Path of Inhaled Nitric Oxide: Biological Consequences of Protein Nitrosylation

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Abstract

Nitric oxide (NO) is a comprehensive regulator of vascular and airway tone. Endogenous NO produced by nitric oxide synthases regulates multiple signaling cascades, including activation of soluble guanylate cyclase to generate cGMP, relaxing smooth muscle cells. Inhaled NO is an established therapy for pulmonary hypertension, especially in neonates, and has been recently proposed for treatment of hypoxic respiratory failure and acute respiratory distress syndrome due to COVID-19. In this review, we summarize the effects of endogenous and exogenous NO on protein S-nitrosylation, which is the selective and reversible covalent attachment of a nitrogen monoxide group to the thiol side chain of cysteine. This post-translational modification targets specific cysteines based on the acid/base sequence of surrounding residues, with significant impacts on protein interactions and function. S-nitrosothiol (SNO) formation is tightly compartmentalized and enzymatically controlled, but also propagated by non-enzymatic transnitrosylation of downstream protein targets. Redox-based nitrosylation and denitrosylation pathways dynamically regulate the equilibrium of SNO-proteins. We review the physiological roles of SNO proteins, including nitroso-hemoglobin and autoregulation of blood flow through hypoxic vasodilation, and pathological effects of nitrosylation including inhibition of critical vasodilator enzymes; and discuss the intersection of NO source and dose with redox environment, in determining the effects of protein nitrosylation.

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In the respiratory tract, nitric oxide (NO) is produced by epithelial cells, inflammatory cells, airway-innervating neurons and vascular endothelial cells¹. It functions as a comprehensive physiological modulator, regulating airway and vascular tone by diffusion from epithelium and endothelium respectively toward smooth muscle; and contributes to immune responses, angiogenesis, hemostasis, neurotransmission, tissue repair and apoptosis². Nitric oxide synthase (NOS) catalyzes the conversion of L-arginine to L-citrulline generating NO, which in turn activates soluble guanylate cyclase (sGC), converting guanosine triphosphate (GTP) to 3',5'-cyclic guanosine monophosphate (cGMP)³.

Inhaled NO (iNO) is an established therapy for pulmonary hypertension, notably for persistent pulmonary hypertension of the newborn (PPHN), based on its primary action of increasing the vasodilator cGMP⁴.

NO is also described as having anti-Coronavirus activity *in vitro*⁵, inhibiting the viral replication cycle by interfering with RNA and protein synthesis⁶. iNO has recently been proposed for treatment of hypoxic respiratory failure in COVID-19, studied in 8 clinical trials currently enrolling patients per www.clinicaltrials.gov, some citing as high as 9 times the inhalational dose regularly used for PPHN treatment. In this context, we will review effects of endogenous as well as exogenous NO on protein S-nitrosylation, a post-translational modification arising from reaction with NO, with wide-ranging effects on protein biological activity.

Nitric Oxide

NO is a lipophilic gas capable of diffusing over a wide area (100 μ m), moving freely across membranes of neighboring cells to function as a short-lived, locally-acting physiological messenger⁷. In biological systems, tissue NO concentration ranges 10-100nM, with a measured half-life of 3-5 seconds⁸. NO signaling can prove beneficial or harmful, depending on duration and local concentration of NO, and the cellular signaling pathways affected. Protein S-nitrosylation (proper chemical term is nitrosation; the term 'nitrosylation' is commonly used to describe the biological function of this modification) is both rapidly reversible and precisely targeted, where NO covalently attaches to the thiol, an organic sulfhydryl group containing a sulfur-hydrogen bond (-SH), on a cysteine amino acid to form a S-nitrosothiol (SNO)^{9,10}, a biologically active intermediate more stable than NO itself¹¹. SNO can arise from endogenous or exogenous NO, from nitrite (exogenous or endogenous, usually derived from dietary nitrate) or other oxidized NO species, metal-NO complexes or protein nitrosothiols (R-SNOs)¹².

NO exists as a free radical with one unpaired electron ($^*\text{NO}$) but without electrical charge. It is not very reactive at physiological concentrations, but can morph into different redox states (Figure 1); NO can lose electrons to become nitroso cation (NO^+) or gain electrons to become nitroxyl anion (NO^-)¹³. NO^+ can react with a thiol group to nitrosylate a protein only if that protein is adjacent; NO^+ will promptly hydrolyze to nitrite in presence of water. In aerobic tissues, NO can be oxidized to nitrite (NO_2^-) or nitrate (NO_3^-) anions, which are less biologically active; or further oxidized to the far more reactive species nitrogen dioxide ($^*\text{NO}_2$), dinitrogen trioxide (N_2O_3), or peroxyntirite (ONOO^-)¹⁴. $^*\text{NO}$ itself reacts slowly with a neutral thiol; but if that thiol is protonated to a thiyl radical (RS^*) by $^*\text{NO}_2$, $^*\text{NO}$ will react with it very rapidly to form S-nitrosothiol¹⁵. This reaction is facilitated by presence of superoxide anion (O_2^-)¹⁶. A reduced thiol (thiolate, RS^-) can be nitrosylated by N_2O_3 formed by oxidation of $^*\text{NO}$ oxidized by oxygen to NO_2 , then recombined with $^*\text{NO}$. Both RS^- and RS^* will react with ONOO^- , toxic product of $^*\text{NO}$ with O_2^- , resulting in S-nitrosothiol⁸. The redox environment around $^*\text{NO}$ becomes critical when $^*\text{NO}$ is in pharmacological concentrations, driving first order reactions that greatly promote S-nitrosothiol formation¹⁴.

Cysteine S-Nitrosylation

Cysteine is among the least abundant residues in proteins, but one of the most conserved¹⁷. It is functionally unique due to its highly reactive thiol side chain, which is nucleophilic, acidic and redox active¹⁸ due to the presence of hybrid orbitals, making electron transfer reactions more likely at the thiol group¹⁹. The frequency of cysteine residues is greater in transmembrane proteins (3.5%) versus purely intracellular proteins (1.6%)²⁰, and more common in extracellular (3.2%) than cytosolic domains (1.0%)²¹. Cysteines play a principal structural role in the formation of disulfide bonds critical for correct protein folding²², promote catalysis by redox enzymes^{23,24}, and regulate protein-protein interactions²⁵.

Cysteine S-nitrosylation serves as a mediator of NO-related bioactivity, both in NOS-containing cells and following NO diffusion for intercellular signaling²⁶. Over 3000 proteins are regulated by nitrosylation, impacting biological processes ranging from DNA damage repair, transcriptional regulation, cell growth, differentiation and apoptosis, to protein stability and redox regulation²⁷. Dysregulated S-nitrosylation is implicated in a number of pathologies²⁸⁻³¹.

Reversible thiol nitrosylation^{32,33} is not the only post-translational modification to which cysteines are susceptible. Others include glutathionylation³⁴, carbonylation³⁵, guanylation³⁶ and sulfhydrylation³⁷. Several of these can alter protein stability, hydrophobicity or protein-protein interactions; but only nitrosylation acts in a rapidly reversible manner suitable for cell signaling³⁸⁻⁴¹. In this, nitrosylation shares similar properties with phosphorylation. Both modifications exhibit substrate specificity, and strict spatiotemporal regulation. However, while the specificity of serine/threonine phosphorylation is determined by kinase enzymes' selectivity for their protein targets, cysteine nitrosylation is unique in being largely non-enzymatic¹⁰.

Nitrosothiols: Nitrosylation Substrates and Donors

S-nitrosylated proteins (R-SNO) are formed by exposure of redox-active motifs to NO, by nitrosyl moieties transferred to the target cysteine from other proteins (termed protein-assisted transnitrosylation), or enzymatically through activity of S-nitrosylase or metalloprotein-catalyzed reactions^{42,43}. In non-enzymatic

transnitrosylation, the nitroso (NO^+) group carried by a donor R-SNO is dropped onto an accessible target cysteine on a nearby protein with the correct surrounding amino acid configuration, creating a new R-SNO.

While 90% of all S-nitroso groups are anchored to cysteine residues of large proteins like albumin or hemoglobin, 10% involve low molecular weight proteins (eg cysteine/S -nitrosocysteine, glutathione/S -nitrosogluthathione) which function as intermediates in transnitrosylation⁴⁴. Transnitrosylation can create a cascade similar to a kinase signaling cascade, whereby one nitrosylation event triggers a next and a next. Not every cysteine in a protein can get S-nitrosylated. Nitrosylation is regulated by close protein interactions, and by the presence of “SNO motifs” marking a particular cysteine as a nitrosylation candidate: a sequence of adjacent amino acids consisting of nucleophilic residues (generally an acid and a base), which may result from the protein’s tertiary or quaternary structure⁴⁵. Proteomic analyses reveal nitrosylated cysteine residues sitting in a hydrophobic pocket, immediately flanked by acidic (Asp, Glu) and basic (Lys, Arg, His) amino acids^{46,47}. The surrounding residues determine cysteine S-nitrosylation, by altering thiol nucleophilicity in that subcellular environment⁴⁵. The presence of charged residues in the vicinity of a cysteine residue provides sites for protein-protein interfaces that direct the S-nitrosylation of that cysteine⁴⁸. This protein micro-environment guides site-specific transnitrosylation; there is little overlap between these nitrosylatable cysteines, and those cysteines targeted by other redox-dependent or enzymatic modifications¹⁰. The propensity for nitrosylation of a given cysteine can be predicted by computational analysis of the surrounding amino acid sequence⁴⁹.

The nature of the R determines R-SNO chemistry, as R-SNOs vary in their stability^{50,51}. CysNO (nitrosocysteine) structure exhibits different characteristics than HSNO (thionitrous acid, the smallest possible R-SNO); S-nitrosogluthathione (GSNO), slightly larger, is more stable^{52,53}. S-nitrosothiols can transnitrosylate thiol-containing amino acids, peptides, or proteins by releasing NO^+ or NO^- , more rapidly than they can spontaneous release NO^{13} . The nitroso group can also move from stable S-nitrosoproteins to less stable species such as CysNO^{54,55}, which in turn can non-enzymatically decompose to cysteine and an NO radical^{54,56}.

Regulators of Denitrosylation: GSNO and Thioredoxin

Turnover of the cellular pool of NO reflects the balance between protein nitrosylation triggered by NO, and the enzymatic denitrosylation of S-nitrosylated proteins^{57,58}. The cellular balance of protein transnitrosylation to denitrosylation can be determined by an equilibrium between glutathione to nitrosogluthathione, thioredoxin to thioredoxin reductase, or other intracellular SNO carriers⁵⁹(Figure 2).

S-nitrosogluthathione (GSNO) is the most physiologically relevant R-SNO in the respiratory system⁶⁰. Its nitrosylation state is regulated by GSNO reductase (GSNOR), an evolutionary conserved, ubiquitous NADH-dependent alcohol dehydrogenase enzyme^{61,62}. GSNOR metabolizes GSNO to oxidized glutathione plus a glutathione N-hydroxysulfenamide intermediate, which is ultimately reduced to ammonia⁴⁴.

Denitrosylation of nitrosylated proteins is catalyzed by reduced glutathione (GSH) which takes the nitroso group to become GSNO, leaving free -SH groups on denitrosylated cysteine residues. GSNO is a substrate for GSNOR, which reverts it GSH⁶¹. Since GSNO is thus in equilibrium with other S-nitrosylated proteins, increased levels of GSNOR indirectly regulates the cellular level of S-nitrosylated proteins, by promoting GSNO denitrosylation⁶³. Pharmacological inhibition or genetic deletion of GSNOR leads to enhanced systemic vasodilation^{64,65}, and GSNOR-null mice have increased blood levels of S-nitrosylated proteins⁴⁴. That GSNOR deletion is associated with decreased systemic vascular resistance is consistent with a proposed role for GSNO in conveying the activity of pulmonary eNOS-derived NO to the systemic circulation¹⁹. On the other side of the equation, an imbalance in the cellular redox state that increases superoxide ion and peroxynitrite promotes S-nitrosylation, by decreasing levels of reduced GSH^{66,67}.

The other major regulator of the cellular protein thiol redox state is thioredoxin (Trx), a ubiquitously conserved oxidoreductase which limits protein nitrosylation^{68,69}. In contrast to GSNOR’s strict substrate specificity for GSNO, cytosolic and mitochondrial thioredoxins directly mediate the denitrosylation of diverse R-SNO protein substrates¹⁹. Thioredoxin’s Cys-X-X-Cys active site⁷⁰ tackles denitrosylation as a multi-

step process: (i) the nucleophilic cysteine displaces NO from the R-SNO's cysteine by heterolytic cleavage, resulting in formation of a mixed disulfide bridge between thioredoxin and its target substrate, (ii) HNO is released, (iii) the second reactive cysteine resolves the disulfide bridge, forming an oxidized variant of thioredoxin and releasing the now reduced substrate, and (iv) the thioredoxin active site is finally reduced by thioredoxin reductase (TrxR)^{71,72}. Target R-SNO proteins can also directly transnitrosylate one of thioredoxin's active site cysteines, causing release of HNO⁷³. Thioredoxin plays an active role in attenuating NO signaling as well as in ameliorating nitrosative stress. NO can reciprocally modulate the redox activity of thioredoxin, by nitrosylating thioredoxin itself⁷⁴.

Endogenous NO and Compartmentalized Nitrosylation

Generation of intracellular NO is tightly linked to oxygen homeostasis⁷⁵, primarily by nitric oxide synthase (NOS) enzymes⁷⁶. The three isoforms of NOS in mammalian cells include neuronal NOS (nNOS; NOS1), inducible NOS (iNOS; NOS2) and endothelial NOS (eNOS; NOS3). Endogenous NO is produced by NOS from its substrate L-arginine^{27,77}. NOS isoform expression differs across organs, tissues, cells, or subcellular compartments. eNOS and nNOS are constitutively expressed, stimulated by intracellular calcium. nNOS is found in neurons, also in kidney, liver, gut and lung⁷⁸. Calcium-insensitive iNOS is induced during the inflammatory response, contributing to the respiratory burst of phagocytes⁷⁹. eNOS is induced in endothelium by exercise and sheer stress⁷⁸, and by chronic hypoxia⁸⁰, which also upregulates eNOS's chaperone heat-shock protein Hsp90, calcium-independently stabilizing the NO output of eNOS⁸¹. Smooth muscle also expresses eNOS, though not generating NO in physiologically important amounts⁸².

At physiological levels of NO, endogenous SNO concentrations are in the low micromolar range^{60,83-85}. Inducible NOS generates higher local concentrations of NO than do other NOS isoforms. A high local NO concentration, in presence of oxygen, is thought to be required for S-nitrosylation of proteins^{86,87}, though lower NO levels may be sufficient where S-nitrosylation involves reactions between NO and thiyl radicals or is catalyzed by transition metals^{88,89}. Each NOS isoform can also be S-nitrosylated, which down-regulates their activity^{90,91}. NO-mediated nitrosylation of eNOS is followed by transnitrosylation of eNOS regulators including Hsp90⁹², inhibiting the activation of eNOS in a negative feedback mechanism limiting production of additional NO.

Though NO is freely diffusible, its concentration due to compartmentalization of NOS creates localized environments conducive to the S-nitrosylation of specific proteins. The physical proximity of SNO targets to NOS favors interaction between NOS enzymes and their nitrosylation substrate, directly or through scaffolding proteins⁹³. In endothelial cells, nitrosylation target proteins are typically found in close proximity to eNOS, to promote their S-nitrosylation⁹⁴; the NO generated by eNOS remains within its subcellular region, confining S-nitrosylation to those proteins adjacent⁹⁵. eNOS is found in the Golgi body, generating a local NO pool that enhances nitrosylation of proteins within the Golgi; and in membrane caveolae where nitrosylation may regulate receptor signaling complexes⁹⁶. S-nitrosylation of proteins in the Golgi reduces granule trafficking from Golgi body to plasma membrane⁹⁶. The eNOS located at the plasmalemma is efficiently activated by calcium-dependent or Akt-dependent agonists, while eNOS on the Golgi body is less responsive to agonists. Consequently, eNOS at membrane caveolae releases a greater amount of NO, compared to eNOS in Golgi^{97,98}. Caveolar eNOS also colocalizes with cortical actin filaments, so eNOS activation can respond to mechanical stimuli such as sheer stress; actin depolymerization or cytoskeletal reorganization can alter eNOS activity and thus alter nitrosylation of caveolar proteins⁹⁹. There is also selectivity of cysteine targeting among NOS isoforms; different NOS isoforms can bind to different regions of the same target protein, resulting in S-nitrosylation of multiple cysteine residues¹⁰⁰. This profound localization of effect in the case of endogenous NO permits the targeted nitrosylation of proteins within subcellular compartments, spatially and temporally associated with NOS activation. Such fine target control is not reported when discussing exogenous sources of NO, which may cause nitrosylation of proteins in many cell types or cell compartments in the course of their pharmacological action.

Nitrates and Nitrites Generating NO

Nitrites elicit physiological responses reminiscent of free NO, including vasodilation and inhibition of platelet aggregation, resulting in their therapeutic use as NO donors. Nitroglycerin, its long acting congener isosorbide dinitrate, and sodium nitroprusside (SNP) are commonly used NO-releasing drugs for treatment of systemic hypertension and coronary artery disease. SNP releases NO rapidly without need of enzymatic catalysis, while nitroglycerin releases NO with the aid of mitochondrial aldehyde dehydrogenase¹⁰¹. Nitrite anion (NO_2^-) is reduced by circulating or tissue bound nitrite oxidoreductases to create a stable, nonreactive pool of NO, and contributing thence to SNO formation. Nitrite reductase is potentiated by hypoxia, so endogenous NO_2^- is promptly reduced to NO if tissue oxygen levels decrease, resulting in protective vasodilation¹⁰².

In adults, the majority of circulating NO_2^- is derived from dietary nitrate (NO_3^-), while in neonates, nutritional sources are poor in NO_3^- ¹⁰³ resulting in lower conversion to NO_2^- ¹⁰⁴ and thus greater dependence on NOS to generate NO. Serum nitrite levels are significantly lower in infants with or without pulmonary hypertension, compared with adults¹⁰⁴. Treatment with inhaled ethyl (alkyl) nitrite, a volatile NO_2^- donor, is an effective acute pulmonary vasodilator in PPHN¹⁰⁵, and prevents hyperoxia-induced lung injury in neonatal rats at least as well as inhaled NO¹⁰⁶. Systemic or inhaled sodium nitrite (NaNO_2) is an alternative NO_2^- based therapy with relatively potent pulmonary vasodilator effect in the setting of experimental models of chronic hypoxic pulmonary hypertension^{107,108}. Nebulized NaNO_2 acutely reduces pulmonary vascular resistance in pulmonary hypertensive patients with β -thalassemia¹⁰⁹. In chronically hypoxic rat pups, subcutaneous NaNO_2 increases tissue NO, ameliorating pulmonary hypertension comparably to inhaled NO, while also increasing R-SNO content in the lung¹¹⁰. NaNO_2 increases lung NO and SNO contents to a greater extent than does inhaled NO, albeit without causing protein nitration¹¹¹⁻¹¹³. Nebulized NO_2^- inhibits hypoxic pulmonary vasoconstriction without decreasing pulmonary arterial pressure or improving oxygenation; in contrast, nebulization of acidified forms of NaNO_2 induce higher levels of NO and thus have greater vasodilator effects than NO_2^- alone¹¹⁴, suggesting it is the NO release and not simply the SNO formation that is the causative agent for NaNO_2 -triggered vasodilation. Nebulization of L-cysteine during treatment with nitric oxide increases formation of circulating S-nitrosocysteines without further increasing pulmonary vasodilation¹¹⁵, supporting the theory that SNO is not itself a pulmonary vasodilator.

Exogenous NO

Exogenous NO is delivered to the lungs by titrated inhalation (iNO), or to the systemic circulation by vasoactive nitrites. Most current treatment regimens use iNO doses up to 20ppm, considered a relatively safe range; clinically apparent toxicity, primarily methemoglobinemia, has been observed at doses above 80ppm. However toxicities at a cellular level, including protein nitration or nitrosylation, are reported at lower doses¹¹⁶. Toxicity is dose-dependent and augmented during oxidative stress. Nitrogen dioxide ($^*\text{NO}_2$), the reactive NO derivative more directly toxic to the lung, can reach a concentration of 5ppm within 12 min during treatment with 20ppm iNO in 100% O_2 , and at 80ppm iNO within 3 min; while in 21% O_2 this reaction takes more than 1 hour¹¹⁷. The use of continuous-flow ventilation circuits has largely eliminated toxic $^*\text{NO}_2$ accumulation within the airways during iNO treatment, though there has been some concern about gas stasis in distal airways during use of high-frequency modes¹¹⁸.

The interaction between the oxidative environment and iNO determines the nature of its downstream nitrosative chemistry (Figure 1). Alone, oxygen toxicity is known to manifest as protein tyrosine nitration, largely due to the action of ONOO^- . Nitrotyrosine can lead to enzyme inactivation and also accelerate protein degradation¹¹⁹. This finding has been well described in hyperoxia models of bronchopulmonary dysplasia¹²⁰. However a shift occurs, when iNO is introduced, in the balance between protein nitration and nitrosylation. Exposure to 80% O_2 rapidly induces both tyrosine nitration and cysteine nitrosylation. Increase to 95% O_2 selectively increases nitration, disfavoring nitrosylation; but addition of 20ppm iNO limits the increase in nitrotyrosine while enhancing formation of S-nitrosocysteine¹²¹. This is ascribed to the excess NO reacting with ONOO^- to produce $^*\text{NO}_2$ and NO_2^- , resulting in decreased ONOO^- available for reaction with tyrosine. If tyrosine nitration were taken as the primary cause of tissue injury, this would imply iNO decreases the toxicity of hyperoxia¹²²; and indeed very high doses of iNO are reported to mitigate the lethality of 95%

O₂ exposure¹²³. Nitrosocysteine toxicity, if present, needs to be accounted for separately. Administration of high dose NO with hyperoxia causes inflammation and surfactant dysfunction, relieved by coadministration of superoxide dismutase, suggesting that nitrosocysteine toxicity is present¹²⁴.

After iNO administration, regardless of concurrent oxygen tension, nitrosylation is the primary protein modification seen in airway and alveoli¹²⁵. An oxygen-rich environment is required for SNO formation; anaerobic solutions of NO are not capable of nitrosylating cysteine or glutathione, but will do so if O₂ is present with NO¹²⁶. NO autooxidation to N₂O₃ follows second order kinetics, but where NO is in pharmacological quantities, this process may be sufficient to promote R-SNO formation¹²⁷. The fate of exogenous NO further depends on the rate of combination of NO with reactive oxygen species. NO reacts readily with O₂⁻ to form ONOO⁻, which will induce thiol nitrosylation; but in vivo the reactivity of ONOO⁻ is dramatically modified by its rapid reaction with CO₂ which abolishes its nitrosative efficiency¹²⁸. Contemporaneous generation of NO and O₂⁻ results in 100-fold more efficient generation of SNO than does ONOO⁻, via a ONOO⁻ independent, CO₂-insensitive mechanism¹²⁹. SNO formation is linear as NO concentration increases; but when the concentration of O₂⁻ approaches equimolar with NO, SNO formation increases exponentially¹²⁷. Higher concentrations of NO inhibit direct lipid peroxidation by ONOO⁻, suggesting NO draws O₂⁻ into other reactions rather than generating ONOO⁻¹³⁰; the preferential formation of SNO is imputed. Neither ONOO⁻ nor O₂⁻ freely diffuse across membranes, as they encounter an energetic barrier at the hydrophobic lipid bilayer, though they can traverse cell membranes via anion channels¹³¹. Under physiological conditions this would effectively compartmentalize SNO formation; but not in presence of an excess of diffusible iNO.

O₂⁻ formation is a hallmark both of hyperoxic and hypoxic pulmonary environments, in the former case formed directly from O₂¹³², and in the latter due to respiratory uncoupling¹³³ and nitrative inhibition of superoxide dismutase¹³⁴. Thiol oxidation increases as hypoxia worsens and cytosolic ROS stress increases¹³⁵. Acute hypoxia is sufficient to increase endothelial cell protein nitrosylation¹³⁶. A decrease in local superoxide dismutase activity decreases bioavailability of NO in the microcirculation¹³⁷. As iNO is usually administered to hypoxemic patients, in combination with therapeutic high oxygen concentrations, it can be assumed that under both circumstances, the reaction kinetics of NO with O₂⁻ described above would drive nitrosothiol formation more than any other post-translational modification.

NO and Hypoxic Vasodilation: Nitrate Reduction or Nitrosylation of Hemoglobin?

Nitrosothiol formation may be viewed as a beneficial modification, largely because of its putative role in mediating hypoxic vasodilation, which couples tissue oxygen delivery to metabolic demand. The circulation of oxygenated hemoglobin to a hypoxic tissue results in dissociation of oxygen from hemoglobin, but also triggers localized vasodilation, increasing blood flow. Two mechanisms have been proposed: that oxygenated blood carries SNO as a NO equivalent, and upon deoxygenation, nitrosohemoglobin releases a nitrosylated intermediate which is exported from red blood cells to the plasma as a low-molecular-weight R-SNO, leading to hypoxic vasodilation¹³⁸; and/or that NO₂⁻ is the circulating NO reservoir, and deoxyhemoglobin acts as a nitrite reductase, producing from NO₂⁻, NO and N₂O₃, which cause vasodilation¹³⁹. Both hypotheses arise from observations of arteriovenous gradients, in case of the SNO-Hb hypothesis a gradient of nitrosohemoglobin, with an increase in plasma S-nitrosothiols in the venous circulation¹⁴⁰; and in case of the nitrate hypothesis, the arterial to venous decline of plasma NO₂⁻, due to its consumption¹⁴¹.

In the SNO-Hb paradigm, when oxygen is bound to hemoglobin molecules in the lungs, the conformational change of the beta subunit permits NO reaction with cysteine, forming S-nitrosohemoglobin. Upon delivery of oxygen to the tissues, deoxygenation of hemoglobin is accompanied by release of the NO group. Dissociation of nitrosohemoglobin results in the measurable formation of R-SNO in distal locations, including airway, gut and extracellular spaces^{33,60,88,142,143}. While most R-SNO are membrane-impermeable, vasodilatory signaling occurs intracellularly. So R-SNOs may decompose to release free NO, either spontaneously or via catalysis by cell surface protein disulfide isomerase^{144,145}, allowing NO diffusion into the cell. Alternatively, circulating R-SNO can transnitrosylate H₂S to generate membrane-permeable HSNO¹⁴⁶; or similarly CysNO, or GSNO hydrolysis to CysNO-glycine, which enter cells via the L-type amino acid transporter^{147,148} or dipeptide transporter¹⁴⁹ respectively. By facilitating transfer of NO in the nitroso form, hemoglobin functions

as a physiological sensor of local oxygen tension, directly modulating local blood flow and oxygen delivery. S-nitrosohemoglobin contracts blood vessels, decreasing cerebral perfusion; however, where the transported NO is released, vessels relax and blood flow improves¹⁵⁰.

In the nitrite paradigm, while the degree of hypoxia and acidosis required for NO_2^- reduction by xanthine oxidoreductase, and the high concentration of NO_2^- required to drive this reaction, initially made NO_2^- an unlikely intrinsic vasodilator, its ready intravascular reduction by deoxyhemoglobin reinstated it as a prime candidate for the transduction of NO vasodilator activity to tissues¹³⁹. Deoxymyoglobin in skeletal muscle also is an abundant nitrite reductase¹⁵¹ while plasma ceruloplasmin acts as a nitrite synthase, oxidizing NO to NO_2^- ¹⁵². As a potential storage pool for NO, there is a lot more nitrite available; circulating levels of NO_2^- range 150-1000nM, while circulating plasma nitrosothiols are estimated under 10nM¹³⁹. This view gained ground after mutation of the conserved Cys93 residue of hemoglobin showed no loss of red blood cell-dependent hypoxic vasodilation¹⁵³. Alleviation of pulmonary hypertension and attenuation of remodeling following nitrite inhalation in hypoxic animal¹⁵⁴ and human subjects¹⁵⁵ support this hypothesis. However an arterial infusion of NO_2^- produces no increase in blood flow, while eNOS activation by acetylcholine dose-dependently increases venous NO_2^- levels while augmenting local blood flow, suggesting the measured NO_2^- may be an indicator of endogenous NO but not itself act as a vasodilator¹⁵⁶.

SNO-Hb and nitrite paradigms are not mutually exclusive hypotheses, but one may predominate in physiological conditions. A recent elegant defense of the SNO-Hb paradigm paints a picture of respiration and blood flow autoregulation as a balancing act of three gases: oxygen, CO_2 and NO¹⁵⁷, where intracellular nitrosothiols transduce the activity of NO to permit red blood cells to autoregulate blood flow for gas exchange. Stored blood is deficient in nitrosohemoglobin, impairing oxygen delivery by transfused red cells; this is restored when its hemoglobin is re-nitrosylated¹⁵⁸. In the pulmonary circulation, SNO-Hb is purported to influence ventilation perfusion matching; exposure to sustained hypoxia causes a build-up of heme-nitrosyl species unable to proceed to the oxygen-requiring step of thiol nitrosylation, resulting in local pulmonary vasoconstriction¹⁵⁹. Experimental inhalation of O-nitrosoethanol, a gas that induces nitrosothiol formation without releasing NO or NO_2^- , restores hemoglobin nitrosylation, and also pulmonary vasodilation¹⁶⁰.

Transnitrosylation of hemoglobin depends on local NO and oxygen concentrations, and the ratio of NO to hemoglobin. In physiological concentrations, NO preferentially nitrosylates hemoglobin thiols or reacts with O_2^- (prevented within red cells by their efficient superoxide dismutase activity), rather than being oxidized to NO_2^- ¹⁶¹. Certain other nitroso donors are possible; among small R-SNOs, CysNO can access the buried cysteine residue in hemoglobin, while GSNO is too large to transnitrosylate hemoglobin¹⁶². NO_2^- can nitrosylate hemoglobin as well⁶². In absence of oxygen, the NO bound to hemoglobin is released as NO^- , resulting in methemoglobin. Under homeostatic O_2 conditions, if the NO:Hb ratio exceeds physiologic, nitrosylation of hemoglobin paradoxically decreases, while methemoglobin increases, particularly of deoxyhemoglobin. At high NO concentrations, NO will reduce methemoglobin to generate NO_2^- , without increasing nitrosohemoglobin¹⁶³. Thus, the effects of endogenous NO and exogenous NO on beneficial hemoglobin nitrosylation cannot be assumed to be identical; more is not always better.

Consequences of Protein Nitrosylation

Endogenous nitrosylation may activate or inhibit protein activity. It is often difficult to discern whether nitrosylation of a specific protein is protective or dangerous, as cumulatively measured SNO does not change while the nitroso group is swapped from protein to protein. S-nitrosylation of thioredoxin is reported to decrease apoptotic signal in endothelial cells³⁹. But a complex dynamic reveals that while cytosolic caspase-3 indeed rests in a state of denitrosylation due to its transnitrosylation of thioredoxin, maintaining homeostasis, mitochondrial caspase-3 denitrosylation, also by means of thioredoxin transnitrosylation, triggers caspase-mediated apoptosis⁷¹. Also in endothelial cells, constitutive S-nitrosylation of NADPH subunit p47phox limits O_2^- production¹⁶⁴; but increased nitrosylation of arginase triggers uncoupling of eNOS and worsens oxidative stress¹⁶⁵. Nitrosylation can alter receptor stability or turnover, altering airway and vascular smooth muscle responses. S-nitrosylation of GRK2 inhibits interaction of GRK2 with adrenergic receptors, preventing receptor phosphorylation and internalization¹⁶⁶. This averts tachyphylaxis and potentiates responses to

adrenergic agonists, both constrictor and dilator. On the other hand, nitrosylation of dynamin promotes receptor internalization¹⁶⁷. Protective effects of endothelial protein nitrosylation include nitrosylation of cell surface tissue plasminogen activator, increasing antiplatelet or fibrinolytic properties¹⁶⁸ and inhibiting neutrophil adherence to the endothelial surface¹⁶⁹. Low dose iNO prevents apoptosis in endothelial cells subject to VEGF receptor blockade¹⁷⁰, and decreases fibrin deposition following hyperoxic injury¹⁷¹.

Excess nitrosylation carries negative consequences. In septic shock, hypotension and organ dysfunction is associated with iNOS induction and widespread protein nitrosylation, while reversal of nitrosylation improves blood pressure and mortality¹⁷². Reduction of cardiomyocyte protein S-nitrosylation by the enzyme GSNOR reverses septic myocardial depression¹⁷³. GSNOR-null mice have a global increase in S-nitrosylated proteins, develop systemic hypotension under anesthesia and have higher mortality in sepsis⁴⁴. During brain reperfusion after hypoxic ischemia, nitrosylation of the survival regulator PTEN increases, worsening brain injury; this is ameliorated by NOS inhibition¹⁷⁴.

Within the lung, iNO in presence or absence of hyperoxia causes rapid protein nitrosylation in airway epithelium and alveolar interstitium, and also in vascular endothelium¹²⁵. In endothelium, eNOS is abruptly inhibited by its own nitrosylation¹⁷⁵, and by nitrosylation of Hsp90, in presence of exogenous NO⁹². While iNO triggers nitrosylation of cysteine to CysNO in alveolar lining fluid¹¹⁵, the transmembrane passage of that CysNO to facilitate transnitrosylation of hemoglobin is mediated by the L-amino acid transporter, which is impaired by hyperoxia, resulting in diminished NO effect¹⁷⁶. CysNO corralled within the alveolar space can cause local damage. Surfactant protein-D, when nitrosylated, becomes a chemoattractant triggering inflammatory signaling¹⁷⁷. Epithelial Na⁺ channels (ENaC), critical for regulating alveolar fluid, are inhibited by nitrosylation causing pulmonary edema¹⁷⁸.

Following vascular smooth muscle challenge with CysNO, proteins identified by mass spectrometry as targets of S-nitrosylation are localized to cytoskeleton and ER/Golgi transport system, regulating protein folding, signal transduction, proliferation and apoptosis⁴⁶. S-nitrosylation inhibits the RhoA pathway, thereby down-regulating myosin phosphorylation¹⁷⁹. However, an important deleterious effect of excess nitrosylation in vascular smooth muscle is guanylate cyclase inhibition (Figure 3). Soluble guanylate cyclase (sGC) acts as the intracellular receptor for NO, generating cGMP from GTP, thereby activating downstream cGMP-dependent protein kinase (PKG) and promoting vasorelaxation. Its GTP catalytic site is a heterodimer of sGC α_1 and β_1 subunits, each containing a heme-NO/O₂ binding domain¹⁸⁰. When NO encounters a free β_1 subunit, it triggers the insertion of a heme moiety, then dissociation of the β subunit from its chaperone Hsp90 and assembly of the sGC α_1/β_1 catalytic unit¹⁸¹. sGC contains over 30 conserved cysteines, of which Cys122 in β subunit, and Cys243, Cys516 in α subunit, are identified by mutational analyses as nitrosylation candidates^{182,183}; nitrosylation of any one of these cysteines desensitizes sGC activity to NO stimulation. sGC nitrosylation is known to occur during iNO therapy¹⁸⁴ and during treatment with nitrates¹⁸⁵, inhibiting sGC catalytic activity and causing NO resistance^{186,187}. sGC nitrosylation is modulated by the thioredoxin system; inhibition of thioredoxin reductase causes increased S-nitrosylation of cellular proteins including sGC¹⁸⁸, resulting in the loss of NO-mediated arterial relaxation¹⁸⁹. This finding is not unique to exogenous NO treatment; vascular smooth muscle in angiotensin-mediated hypertension¹⁸³ and asthmatic bronchial smooth muscle¹⁹⁰, both conditions marked by increased endogenous NO, are complicated by a nitrosylated, desensitized sGC, unresponsive to NO. Therapeutic developments to reactivate a nitrosylation-damaged sGC include cinaciguat (BAY 58-2667)¹⁹¹ and BAY 60-2770¹⁹², which act as heme-mimetics to restore NO-sensitivity of the sGC β subunit; or riociguat (BAY 63-2521)¹⁹³ and BAY 41-2272¹⁸⁶ that both enhance the sGC response to endogenous NO, and confer NO-independent allosteric activation of sGC. We have reported nitrosylation causing inhibition of the congener vasodilator, adenylyl cyclase, in pulmonary artery smooth muscle following hypoxia, and following exposure to CysNO¹⁹⁴; candidate cysteine residues for nitrosylation are found at the activation interface of adenylyl cyclase with G α s¹⁹⁵. It is possible that excess cysteine nitrosylation affects both major smooth muscle relaxant pathways by similar mechanisms, interfering with enzyme activation.

In sum, cysteine nitrosylation is a homeostatic amplifier of NO-mediated vasodilation, a protector from toxic

protein nitration, and also a threat to alveolar integrity and NO-mediated vasodilation, about which one cannot be cavalier. When treating hypoxic respiratory failure with therapeutic hyperoxia and inhaled NO, side effects involving protein nitrosylation, notably in alveoli and bronchial or pulmonary arterial smooth muscle, can seriously interfere with treatment effects. As higher doses of NO are used, and as we learn more about the complex redox environment that governs localized nitroso group formation, more research is required on the boundaries between beneficial protein nitrosylation, and excess or hazardous nitrosylation.

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Figure Legends

Figure 1: Reactive species generated from nitric oxide, and post-translational modifications

NO can gain electrons to become nitroxyl anion or lose electrons to become nitroso cation; oxidize to nitrite or nitrate; or further oxidize to nitrogen dioxide or dinitrogen trioxide. Cysteines sensitive to nitrosylation are flanked by “SNO motifs” containing basic and acidic residues. Nitroso ion, nitrite, and NO when in presence of superoxide, can react spontaneously with cysteine thiols to form nitrosocysteine. Nitrogen dioxide reacts with cysteine to produce thiyl radical, which reacts with peroxynitrite to make nitrosocysteine. In a competing reaction, peroxynitrite reacts with the aromatic ring of tyrosine to form 3-nitrotyrosine.

Figure 2: Transnitrosylation and denitrosylation mechanisms of protein S-nitrosothiols (R-SNO)

Nitric oxide synthase (NOS) produces NO [A], resulting in auto-nitrosylation of NOS. NO oxidizes to nitrogen oxide (NOx) species including NO₂ and N₂O₃ that can convert glutathione (GSH) to S-nitrosoglutathione (GSNO). [B] GSNO can be generated via the transition metal pathway, where NO reacts with metal-containing proteins to form R-SNOH; the thiyl radical pathway, where NO₂ reacts with a thiol to produce a thiyl radical (R-S•); the NO oxidation pathway where N₂O₃ reacts directly with a thiol to form R-SNO; or by transnitrosation from R-SNO, such as the nitrosylated NOS. [C] Denitrosylation of GSNO leading to formation of non-bioavailable NO requires its reduction by nitrosoglutathione reductase (GSNOR) to intermediate GSNHOH, which reacts with GSH to produce glutathione disulfide (GSSG) and NH₂OH, later reduced by glutathione reductase (GR) to remake GSH; or GSNHOH hydrolysis to GSO₂H and ammonia. [D] Denitrosylation of any other R-SNO requires transnitrosylation of thioredoxin (Trx) to one of its acceptor cysteines (X indicating cysteine position). Thioredoxin is then restored by thioredoxin reductase (TrxR).

Figure 3: Nitric oxide sources, hypoxic vasodilation and vasoconstriction

Nitrite (NO₂⁻) can be endogenous or arise from nitrite donors such as ethyl nitrite (C₂H₅NO₂) or other RNO₂, sodium nitrite (NaNO₂) or nitrate; it is reduced to NO by xanthine oxidoreductase (XOR). In-

haled nitric oxide (iNO) is an exogenous source of NO. NO generates nitrosothiols (SNO) proteins directly through oxidation and conversion, by rapid combination with superoxide (O_2^-), or through NO reaction with superoxide to form peroxynitrite (ONOO⁻). Vasodilator effects of SNO formation include nitrosylation of hemoglobin (Hb) to form SNO-Hb (nitrosohemoglobin), delivering NO to the distal circulation. Glutathione (GSH) nitrosylation can deliver SNO locally to transnitrosylate other proteins. Soluble guanylate cyclase (sGC) is the intracellular receptor of NO, generating cyclic GMP important for arterial relaxation, particularly in the pulmonary circuit. Vasoconstrictor effects of SNO formation include nitrosylation of sGC, inhibiting vasodilation.

Figure 1

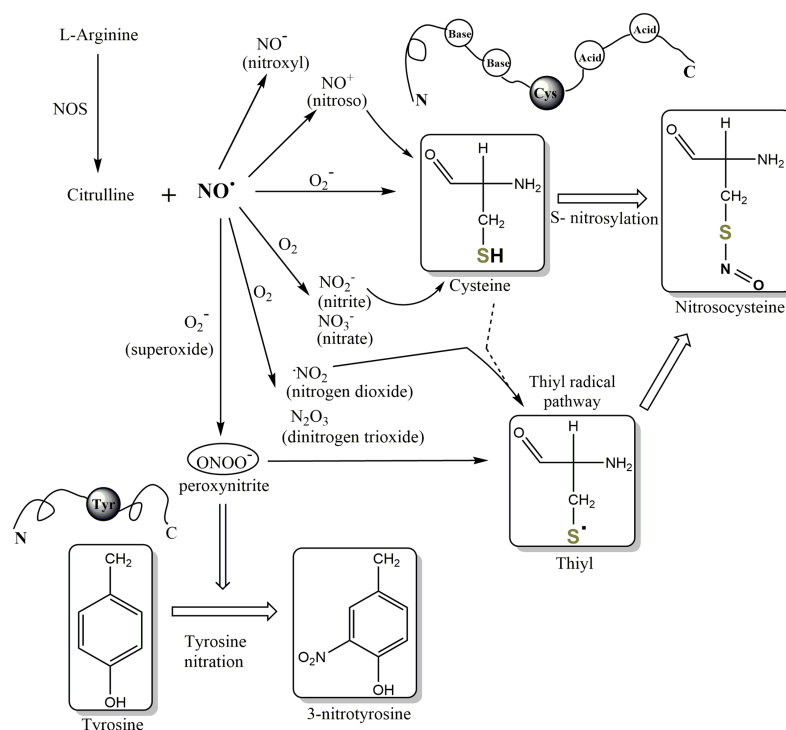


Figure 2

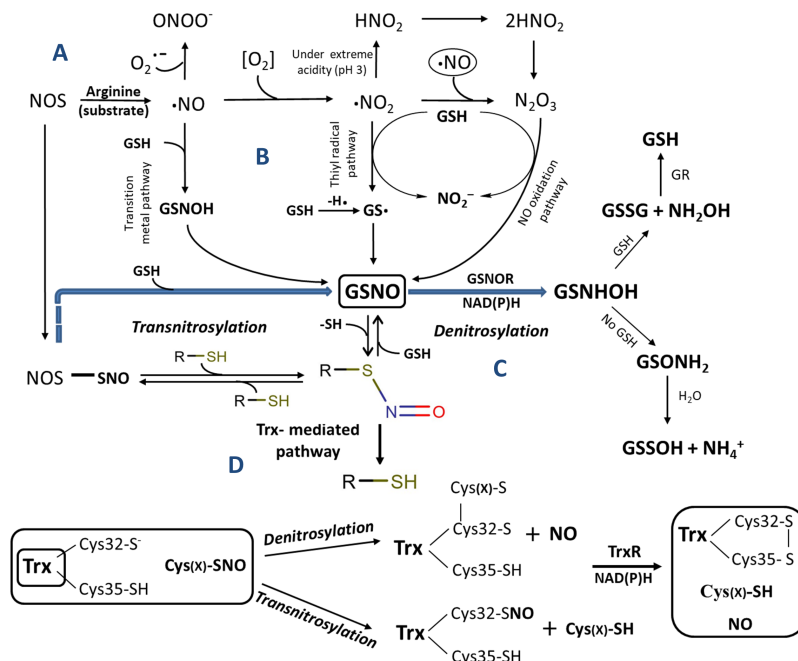


Figure 3

