Warming drives sustained plant phosphorus demand in a humid tropical forest

Zhiyang Lie¹, Wenjuan Huang², kadowaki Kohmei³, Guoyi Zhou¹, Junhua Yan¹, Josep Penuelas⁴, Jordi Sardans⁵, David T. Tissue⁶, Yuelin Li¹, Shizhong Liu¹, Guowei Chu¹, and Juxiu Liu¹

¹South China Botanical Garden ²Iowa State University ³Kyoto University ⁴CSIC-CREAF ⁵CREAF ⁶Western Sydney University

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Abstract

Phosphorus (P) is often one of the most limiting nutrients in highly weathered soils of humid tropical forests, which may regulate the responses of carbon (C) feedback to climate warming. Based on a 7-year continuous field warming experiment conducted by translocating microcosm forest ecosystems from a high-elevation site to low-elevation sites, we detected changes in the ecosystem P cycle in response to warming. We report that warming drives sustained plant P demand by increasing P uptake and thus decreasing foliar N:P. This increased plant P content is supplied by multiple processes including enhanced plant P resorption, soil P mineralization and dissolution without changing litter P mineralization and leachate P. These findings suggest that warming may alleviate initial P deficiency and/or limitation of plant growth and contribute to sustaining plant C fixation in these tropical forests.

Introduction

Tropical forests account for approximately one-third of global terrestrial gross primary productivity and sequester one-half of global carbon (C) stored in terrestrial vegetation (Hubau et al., 2020). Climate models predict that tropics will experience earlier climate warming compared with other climatic regions (Mora et al., 2013). Consequently, large amounts of C stored in tropical forests may be lost by the warming induced release of carbon dioxide from soil due to enhanced microbial activity (Nottingham et al., 2019, 2020). However, it remains uncertain whether tropical forests will increase C storage above-ground in plant biomass to potentially compensate for C lost from soils under climate warming (Reed et al., 2012; Wieder et al., 2015). Phosphorus (P) status could be one of the most important factors determining C balance of tropical ecosystems under warming, as P often limits net primary production in the tropics (Cleveland et al., 2011; Du et al., 2020). If plants experience greater P limitation due to warming, then plant productivity and C fixation might be significantly reduced (Sullivan et al., 2020). Currently, incomplete understanding of P dynamics in tropical forests in response to warming is a challenge for modeling the direction and magnitude of tropical C feedback in future climates (Reed et al., 2015).

The climate-demand/supply hypothesis holds that P limitation can be caused by high phosphorus demand and low phosphorus supply with climate change (Hou et al., 2021). Numerous studies have addressed plant P dynamics in response to warming (Sardans et al., 2006; Hood et al., 2018), and observed an elevated plant P demand due to increased plant production and P uptake (Rui et al., 2012; Diffenbaugh et al., 2013). To assess the potential for soil P supply under warming, most previous studies have focused on soil P availability (Sardans et al., 2006; Dijkstra et al. 2012). For instance, a decrease in plant-available P derived from the limited soil P mobility is observed with warming-induced drying soil (Dijkstra et al., 2012; Yuan & Chen, 2015; Tian et al., 2018). Based on demand/supply hypothesis, enhanced plant P demand, but decreased soil P availability with warming may increase plant and microbial P deficiency (Jonasson et al., 2004). However, in P-deficient tropical ecosystems, focusing only on available P may lead to uncertainties regarding actual soil P supply since changes in available P can be masked by the balance of rapid soil P release and plants and/or microbes P uptake. These uncertain of P supply constrain our understanding on the status of plant P demand relative to P supply.

Combining key processes of P conversion and P fractions allows the assessment of P supply. The current paradigm states that tropical P cycling is dominated by P mineralization to meet rapid P turnover and plant demand (McGroddy et al., 2008; Turner et al., 2018). In general, warming tends to increase organic P mineralization by stimulating soil phosphatase production in optimal moisture environments (Zhou et al., 2013; Gong et al., 2015; Zi et al., 2018; Zuccarini et al., 2020), which further satisfies P demand during accelerated plant growth (Sardans et al., 2006; Rui et al., 2012). However, these studies ignore a important P supply process: plant P resorption that is a crucial strategy for plants to improve P use efficiency and reduce the dependence of plant growth on soil P availability (Vitousek et al., 1982; Yuan & Chen, 2009). In resorption theory, the magnitude of plant P resorption is considered to be a trade-off with P derived from soil, especially from P mineralization (Vergutz et al., 2012; Gerdol et al., 2019). For instance, plants might retain adequate P concentrations in foliar tissues through increased plant P resorption when soil P availability is insufficient under warming (Yuan & Chen, 2015; Zong et al., 2018). Consequently, in tropical forests with high plant productivity and low P availability in weathered soils, plant P resorption may play a critical role in P supply under warming. In addition, recent research found a previously unrecognized pathway of P supply: P bound to iron (Fe) oxides and hydroxides could be dissolved and released during redox fluctuations in humid tropical soils due to the enhanced metabolism and oxygen depletion of microbes and subsequently Fe reduction (Peretyazhko & Sposito, 2005; Chacón et al., 2006; Gross et al., 2020). A reduced redox potential associated with P release from redox-sensitive dissimilatory reduction of Fe oxide-bound P was observed in a wetland ecosystem (Zhang et al., 2015). Although a large fraction of soil P may be bound to redox-sensitive Fe minerals in a humid tropical forest, significant portions of P might be released from Fe reduction due to increased microbial activity under warming (Li et al., 2016). Moreover, the loss of P also affects P supply, but most studies ignore this fact, which prevents an accurate assessment of P cycling.

Most short-term warming cannot accurately anticipate P cycling dynamics at longer term (> 5 years), although from demand/supply hypothesis, tropical plants may be prone to experience more P limitation in long-term warming as P is continuously depleted with plant uptake and leaching. Lack of long-term monitoring in tropical forests indicates our incomplete understanding of the mechanisms controlling P cycling under long-term warming. To address these knowledge gaps, we conducted a 7-year field warming experiment in a humid tropical forest using natural temperature differences across elevations. We assessed P status in plant, litter, soil, and leachate as well as soil acid phosphatase and soil Fe components in the dry and wet seasons. Our previous studies found a continuous increase in plant growth and the accelerated decomposition of soil organic matter due to warming (Fang et al., 2020; Lie et al., 2021). Hence, we hypothesized that warming would increase plant P demand with enhanced plant growth, and P supply would be affected by multiple biological and geochemical processes which are strongly dependent on temperature.

Materials and Methods

Site and treatments

The study was conducted at the Dinghushan Biosphere Reserve in southern China (112°10' E, 23deg10'N).

The reserve is characterized by a tropical monsoon humid climate with a mean annual temperature of 21 degC and mean relative humidity of 80%. The mean annual precipitation of 1900 mm has a distinct seasonal pattern, with about 80% of it falling from April to September (warmer wet season) and 20% occurring from October to March (cooler dry season).

The experimental warming site was constructed by translocating the microcosm forest ecosystem from the high-elevation site (control: 600 m) to low-elevation sites (+1.0degC treatment: 300 m; +2.1degC treatment: 30 m) (Fig. S1). In May 2012, three plots (open-top chambers; 3 m length x 3 m width x 0.8 m depth) were established at each of the elevation sites, representing three replicates of each temperature treatment. Each belowground plot was surrounded by concrete brick wall bonding with ceramic tile, with one hole connected with PVC tubing at the bottom and the top of the chamber wall to collect underground water and surface runoff, respectively. All plots were separated by a 3 m buffer zone. Before planting, we collected soil from each of the three horizons (0-20, 20–40, and 40-70 cm) in the adjacent naturally occurring mountain evergreen broad-leaved forest, and homogenized each layer separately, and then placed each soil horizon into each plot. Six one-year old dominant plant species, including arbuscular mycorrhiza-forming species (*Schima superba*, *Itea chinensis*, *Machilus breviflora*, *Myrsine seguinii*, and *Ardisia lindleyana*) and only one ectomycorrhiza-forming species (*Syzygium rehderianum*) at an elevation of 600 m in the natural forest, were collected and then randomly planted in each plot (six individuals per species). Four 0.2 m x 0.2 m litter traps were placed randomly in each plot. One tree per species was randomly harvested in December 2014, December 2015, and June 2018 to avoid crowding in the chambers.

Soil temperature and soil volumetric water content at 5 cm soil depth were measured in each plot using Campbell 109 constantan-copper thermocouples and time-domain reflectometer probes (CS616, Campbell, USA), respectively. Data were recorded every 30 min using Campbell Scientific (Logan, UT, USA) CR1000 data loggers. Further detailed information regarding the experimental site was described in Lie et al., (2021).

Sample collection and processing

Twenty to thirty leaves were randomly collected per species (S. superba, S. rehderianum, I. chinensis and M. breviflora; the remaining two species were excluded because of their small biomass with less than 1.1% of the total biomass) for each plot in the dry season and wet seasons from 2013–2019. In 2014, 2015, and 2017–2019, monthly litter samples in each plastic net were collected. In December 2017 and 2018 and June 2018 and 2019, soil samples were collected with five replicates per soil depth (0–10, 10–20, 20–40 cm depth) for each plot using a 3.5-cm-diameter stainless steel soil corer and then were pooled by each plot. Soil bulk density was measured using the cutting-ring method. Soil samples were sieved through a 2-mm mesh and then stored at 4degC. All plant samples were oven-dried at 60 degC for at least 24 h before being weighed. Soil pH was measured at a soil-to-water ratio of 1:2.5.

P status in plant, litter, soil and leachate

The P concentrations of plant and litter were measured photometrically after digestion with H_2SO_4 - H_2O_2 (Anderson & Ingram, 1989). Total plant P uptake was calculated by the sums of tissue biomass multiplied by corresponding P concentrations of the four species. Soil microbial biomass P were analyzed using the fumigation-extraction method (Vance et al., 1987): a subsample was extracted using 0.03 M NH₄F and 0.025 M HCl to determine extracted P, and another subsample was fumigated for 24 h with chloroform and extracted with the same extract. The extracted P concentrations were corrected for the sorption of P to soil particles during extraction based on the difference between samples spiked and not spiked with P (25 µg of inorganic P added as KH₂PO₄). The microbial biomass P values were calculated as the differences between the values of fumigated and un-fumigated subsamples corrected by the conversion factor (i.e., 0.4).

Soil P fractionation was measured according to the method described by Tiessen & Moir (1993) and McDowell & Condron (2000). The P in soil was sequentially extracted by 1.0 M NH₄Cl (soluble P), 0.1 M NH₄F (aluminum phosphate, Al-P), 0.1 M NaOH (ferric phosphate, Fe-P). The non-extracted P in soils after sequential extraction was residual P (occluded P). In brief, 3.0 g of finely ground air-dried soil was extracted by 30 mL of each extractant solution on a horizontal shaker. The shaking time was 30 min for NH₄Cl, 16

h for NH_4F and NaOH. The solution after shaking was centrifuged at 10,000 r/min for 10 min to collect supernatant for inorganic and total P analysis. Phosphorus extracted by NH_4Cl was classified as easilyavailable P, while P in NH_4F and NaOH extracts was classified as moderately-available P. The organic P concentration in NH_4F and NaOH extracts were calculated as the difference between total P analyzed after persulphate digestion and inorganic P. The moderately-available organic P was the sums of organic P extracted by NH_4F and NaOH. All soil extracted P was determined using an inductively coupled plasma optical emission spectrometer (ICP-OES; Optima 2000 DV, Perkin Elmer, USA).

Leachate P concentrations were measured colorimetrically. Total leachate P was calculated by summing both surface and underground soil solutions, after multiplying P concentrations by the leachate volume, in each month. Rainfall was similar among treatments throughout the experiment due to similar slope aspects and close distance across the three sites.

Acid phosphates activity and mycorrhizal fungi

We used colorimetric methods to analyze acid phosphates activity (Tabatabai, 1994). Specifically, four grams of moist soil or 0.2 grams of litter was added with 60 ml of 50 mM acetate buffer (pH 5) in a blender for 1 min. A subsample (0.75 ml) of the slurry was then added to 0.75-ml substrate (5 mM). Two controls (buffer + substrate and buffer + slurry subsample) were also analyzed. Each analysis was repeated three times. After 1h incubation at 20 °C in the dark, each sample was centrifuged at $2091 \times g$, then 0.2 ml of supernatant was transferred into a 96-well plate. Finally, 0.03 ml ultrapure water and 0.02 ml 1 M NaOH were added to each well. Absorbance of each well was measured at 405 nm by Multiskan EX (Thermo Scientific, Waltham, MA). A p-nitrophenol (p -NP) calibration curve was performed, and results were expressed in µmol of p -NP released h⁻¹ g⁻¹ of dry soil.

PLFA assays were performed using established methods to estimate the abundance of arbuscular mycorrhizal fungi ($16:1\omega5c$) and ectomycorrhizal fungi ($18:2\omega6$)(65). In brief, lipids were extracted from 8 g of lyophilized soil, fractionated and methylated. The resulting fatty acid methyl esters (FAMEs) were separated using a GC gas chromatograph (Agilent GC7890A; Agilent Technologies Inc.). Peaks were identified using the Sherlock System (v. 6.1; MIDI Inc.).

Soil Fe minerals

Two assays were used to quantify redox-sensitive soil Fe pools (Liptzin & Silver, 2009; Huang & Hall, 2017). First, we used a hydrochloric acid (HCl) extraction to solubilize reactive Fe(II) and total solubilize reactive Fe_{HCl} [Fe(II) + Fe(III)] from short-range order and organo-Fe complexes; 3.75 g fresh soil was extracted in 30 mL 0.5 M HCl solution. The concentrations of Fe(II) and Fe(III) were analyzed colorimetrically. Second, we used citrate-ascorbate to extract chelatable short-range order (oxy) hydroxides and organo-Fe complexes (Fe_{ca}). This pool is generally defined as reducible amorphous or poorly crystalline Fe (Hall & Silver, 2015). To estimate Fe_{ca}, 1.25 g fresh soil subsamples were added to the freshly prepared citrate-ascorbate solution in a 1:30 ratio, vortexed, and then shaken in the dark for 16 h. Then, the samples were centrifuged at 10,000 r/min for 10 min and the supernatant solution was measured using ICP-OES. Higher Fe(II)_{HCl}: Fe_{HCl}and Fe_{ca} can be good indicators of reducing conditions in humid tropical forest soils (Liptzin & Silver, 2015). To determine total free Fe (Fe_{cd}), Fe was measured in citrate-dithionite extractions at a 1:60 dry soil-to-solution ratio.

Statistical Analyses

Whole plot average plant biomass, foliar P concentration and N:P ratio were calculated using a weighting factor for each species, which was estimated by determining the average basal diameter of each species from 2013 to 2019. Foliar N concentrations were originally presented in Lie et al., (2021). Plant P uptake was estimated by the weighted sum of P accumulated by the four species (data from Wu et al., 2020). Plant P resorption efficiency was calculated as $(1 - P_1/P_f \times MLCF) \times 100\%$, where P_1 is litter P concentrations and P_f is the weighted foliar P concentrations of the four species; MLCF is the mass loss correction factor, which is 0.78 for evergreen broad-leaved forest tree species (Vergutz et al., 2012).

Differences in P status in plant, soil and leachate, as well as soil phosphatase activity and soil Fe minerals under treatment were analyzed using linear mixed-effects model. Warming treatment and experimental time were treated as fixed factors, while plot was treated as a random factor. Pairwise comparisons of means were tested by Least Significant Difference test. Data were transformed to meet the assumptions of normality and homogeneity of variances when necessary. Statistical significance was defined as P < 0.05. Linear mixed models were performed by "lmer" function in "lme4" package (Bates et al., 2015) using R version 3.6.1.

Results

Soil temperature, moisture, pH, redox potential, and bulk density

Throughout the experiment of 2013–2019, the average monthly soil temperatures at 5 cm depth were significantly increased by $1.0\pm0.1^{\circ}$ C and $2.1\pm0.1^{\circ}$ C for the mid-elevation site and the low-elevation treatments (hereafter we refer to them as $\pm 1.0^{\circ}$ C treatment and $\pm 2.1^{\circ}$ C treatment respectively) relative to the control (19.7±0.6°C on average; P < 0.01; Fig. S2a). Compared with the control (18.9±0.3% v/v), the average monthly soil moisture at 5-cm depth was significantly decreased in the $\pm 2.1^{\circ}$ C treatment ($\pm 3.8\%$ v/v; P < 0.01), but was not changed in the $\pm 1.0^{\circ}$ C treatment across the whole experiment (Fig. S2b). In-site soil redox potential at 10-cm depth was decreased by 39.1 ± 24.9 mV in the $\pm 2.1^{\circ}$ C treatment during 2017-2019 (P < 0.05), with no interaction between warming treatments and experimental times (Table S1). Warming had no significant effects on soil pH and soil bulk density in the topsoil (0–10 cm).

Plant P status

During the period of 2013–2019, foliar N:P at ecosystem level in the control averaged 23.8 ± 0.6 . In each treatment, foliar N:P was negatively correlated with foliar P concentration but was uncorrelated with foliar N concentration (Fig. S3). In the +2.1°C treatment, foliar N:P was significantly decreased by $4.7\pm1.5\%$ without changing foliar P concentrations throughout the experiment (P < 0.05; Fig. 1a, b). The +2.1°C treatment significantly increased P concentrations by $23.4\pm4.3\%$ in stems, $14.4\pm5.0\%$ in fine roots and $32.4\pm6.2\%$ in coarse roots in 2018 (P < 0.05), but it did not affect tissue P concentrations in 2015 (Fig. S4a). The +2.1°C treatment significantly increased fine roots and decreased abundance of arbuscular mycorrhizal fungi in 2015, but it had no effect on these in 2018 (Fig. S4b, c).

Plant P uptake was significantly increased in both 2015 and 2018 under warming $(33.8\pm7.4\%)$ in the $\pm1.0^{\circ}$ C treatment and $90.6\pm11.8\%$ in the $\pm2.1^{\circ}$ C treatment on average; Fig. 1c). During 2014–2015 and 2017–2019, plant P resorption efficiency was significantly increased by $5.8\pm2.9\%$ in the $\pm2.1^{\circ}$ C treatment relative to the control ($56.8\pm2.3\%$ on average) (Fig. 1d). Plant resorption was negatively correlated to foliar P concentrations, soil NH₄Cl-P concentrations, and soil moisture (Fig. S5). Litter C:P was significantly increased by $26.0\pm9.7\%$ in the $\pm2.1^{\circ}$ C treatment during 2017–2019 (P < 0.01; Table S2). Litter P stock and litter phosphatase activity were not changed under warming during 2017–2019.

P status in soil and leachate

In the control, relative P components were Fe-Pi > Fe-Po > Al-Pi, Al-Po > soluble P in the three layers of soils, and Fe-Pi accounted for at least 50% of easily-available P and moderately-available P components (soluble P + Al-Pi + Al-Po + Fe-Pi + Fe-Po; Table S3, Fig. 2). With both warming treatments, the soluble P concentrations and microbial biomass P were not changed (Table S3). However, the +2.1°C treatment significantly decreased moderately-available organic P concentrations in 0–10 cm soil depth by $15.5\pm7.3\%$ (P < 0.05) relative to the control ($28.4\pm1.5 \text{ mg kg}^{-1}$; Fig. 2a). The Fe-Pi concentrations were significantly decreased by +2.1°C treatment in 0–10 cm soil depth by $6.8\pm2.7\%$ relative to the control ($34.0\pm0.8 \text{ mg kg}^{-1}$), while Al-Pi concentrations were not changed by warming (Fig. 2b, Table S3). There was no significant response of moderately-available organic P and Fe-Pi to warming in the deeper soils (10-20 cm and 20-40 cm). Soil potential acid phosphatase activity in all three depths was significantly increased in the +2.1°C

Averaged monthly total leachate P in the control was 0.02 kg hm⁻² month⁻¹ during 2013–2019 (Fig. 4). Warming had no significant effect on total leachate P throughout the experiment. However, relative to the control, leachate P of surface runoff was decreased by $55.0\pm37.4\%$ in the +2.1°C treatment, while leachate P in underground water was increased by $+21.4\pm9.2\%$ and $36.8\pm11.2\%$ in the +1.0°C and +2.1°C treatments, respectively. There was no significant interaction between warming treatments and times.

Soil Fe minerals

During 2017–2019, The +2.1°C treatment significantly increased Fe(II)_{HCl}:Fe_{HCl} in three depths (P < 0.05). Soil Fe_{ca} concentrations were significantly elevated in soil depth of 0–10 cm (P < 0.05), but were not changed in the deeper soils (10–20 and 20–40 cm) in the +2.1°C treatment. There was no significant difference of Fe_{cd} concentration between control and warming treatments (Table S3). There were no significant interactions between warming treatments and seasons on Fe(II)_{HCl}:Fe_{HCl}. Fe_{ca} and Fe_{cd} concentrations.

Discussion

We examined P dynamics in the plant, soil and leachate in a humid tropical forest exposed to 7-year of continuous field warming. Contrary to most studies which show that short-term warming increases ecosystem P deficiency in arid ecosystems or large scales (Jonasson et al., 2004; Yuan & Chen, 2015), we found that warming by 2.1°C increased plant P uptake and decreased foliar N:P, indicating warming drove sustained plant P demand and may alleviate P deficiency and/or limitation of enhanced plant growth and contribute to sustaining plant C fixation in these tropical forests. It is worth pointing out that the sustained plant P demand could not be explained by traditional views in accessing P acquisition (through fine root and mycorrhizal fungi; Rosling et al., 2016) and supply (through P availability), because neither fine root biomass nor arbuscular mycorrhizal fungi abundance were affected by 6-year warming as they were by short-term warming (Lie et al., 2021), and soil soluble P had no response to warming.

We observed a enhanced P redistribution capacity and soil P mobilization to support increasing plant P demand without altering litter P mineralization and leachate P, which led to a net transfer of P from soil to plant regulated by biogeochemical processes (Fig. 5): (1) warming increased plant P resorption by 4.7%, which enhanced plant P recycling and soil P inputs; (2) warming elevated soil P mineralization by 15.4% and dissolution by 6.8% in the topsoil, thereby improving soil P supply; (3) reduced P loss in surface runoff under warming facilitated P retention for topsoil and shallow roots. Our findings improved the understanding of biogeochemical controls on plant P acquisition under warming, which are crucial to project the future C sequestration potential of tropical ecosystems that are mostly limited by P (Fig.1; foliar N:P > 20) (Güsewell, 2004; Cleveland et al., 2013).

Plant P resorption increased under warming, similar to observations at the global scale that plant P resorption was positively correlated with temperature (Yuan & Chen, 2009; You et al., 2018). Plant growth periods strongly drive plant nutrient resorption (Sun et al., 2016). A warming experiment conducted in subtropical forests showed that foliar growth periods were extended by an average of 5–18 days at 2°C of warming (Gunderson et al., 2013). Thus, warming potentially delays foliar senescence (Chung et al., 2013) and consequently increases foliar plant P resorption (Estiarte & Peñuelas, 2015). Moreover, it is noted that plant P resorption increased with P supply derived from soil (i.e.,enhanced P mineralization and dissolution) under warming, which contradicts the view on a trade-off between nutrient resorption and soil nutrient supply (Vergutz et al., 2012; Gerdol et al., 2019). We found that increased P resorption was likely driven by warming-induced drying soil (Fig. S5), which is consistent with a study in a dry ecosystem (Ren et al., 2018). P resorption may be limited in the capacity to take up P from litter in drier soil. Thus, an increase in plant P resorption may be a more economical P acquisition strategy than increased investment in roots. These findings extend the application of resorption theory to P cycling, and also highlight the key P source from resorption for plants under warming.

Most studies reported that P mineralization in humid forest soils was enhanced under warming (Rui et al., 2012; Zhou et al., 2013; Zi et al., 2018; Zuccarini et al., 2020). Consistent with these studies, warming elevated soil P mineralization by increasing soil potential acid phosphatase activity and decreasing moderately-available organic P in topsoil. The elevated soil P mineralization could be driven by joint increases in plant demand for P and microbial activity (e.g., higher soil respiration (Li et al., 2016)). The increase in soil acid phosphatase activity could be stimulated by microbes to compensate for the reduction of P diffusion in warming-induced drying soil (Allison et al., 2010) and the limitation of microbial P utilization during litter decomposition with higher litter C:P (Table S3; Sardans et al., 2006). Our data thus showed that more P was retained in plants than returned to soil in each litter cycle under warming, which led to a net transfer of P from soil moderately-available organic P to the plant.

At our site, a large fraction of the moderately-available P in the soils was in inorganic form (>58%), suggesting that inorganic P dissolution may be especially important for P supply. In line with our hypothesis, warming enhanced P dissolution, as evidenced by lower Fe-Pi in the topsoil. However, the Al-Pi and pH in the three soil depths were not affected by warming, suggesting that the decrease in Fe-Pi was not caused by soil pH, which served as a critical factor of P dissolution (Navratil, et al., 2009; Devau et al., 2011). Rather, the lower Fe-Pi was likely related to sensitive changes of soil Fe phases under warming. In topsoil, warming led to decreases in redox potential and increases in both $Fe(II)_{HCl}$: Fe_{HCl} and Fe_{ca} concentrations, indicating higher Fe reduction. Our finding challenges the prevailing expectation that warming drives more oxidative soil, due to increased soil porosity by warming-induced drying soil (Wang et al., 2017). The higher Fe reduction probably was due to the enhanced microbial activity, root respiration coupled to higher P uptake capacity and thus oxygen depletion under warming at our site (Li et al., 2016). Compared with Al minerals, the reduction of Fe minerals is considered to effectively reduce geochemical sequestration of easily-available P to increase soluble P and even alleviate P limitation in humid tropical soils (Chacón et al., 2006; Lin et al., 2018; Gross et al., 2020). Some studies in subtropical wetland systems have reported similar declines in redox potential and increases in P release from Fe minerals under warming (Zhang et al., 2012; Wang et al., 2013). Accordingly, the increases in P release can elevate soil soluble P, as observed in incubation experiments with higher soluble P under reducing conditions (Peretyazhko & Sposito, 2005; Wang et al., 2017). However, we did not find changes in soil soluble P, which might suggest a rapidly attained equilibrium, in which soluble P induced from the dissolution of Fe-Pi was rapidly taken up by plants in this P-poor system (Liptzin & Silver. 2009). The rapid plant uptake could prevent potential P re-immobilization and loss by leaching. Overall, the increase in P release from Fe minerals provides evidence for greater soil P supply under warming, which was expected to facilitate plant P utilization.

The losses of P deserve to be noticed under warming. P in terrestrial ecosystems is primarily lost through leachate that makes it difficult its return to the system. Currently, global P mineable reserves are gradually being lost from ecosystems due to improper use of P fertilizers in forest and farmland managements. Greater plant P uptake generally promotes P fixation in plants and thereby potentially reduces P loss. Intriguingly, we noted that the increased plant P uptake does not imply less P losses, because the net P losses were counterbalanced by the decreased P losses in surface runoff and the increased P losses in underground water. These results suggest that warming shifts P loss from surface to subsurface, and emphasizes the necessity to assess effects of warming on P dynamics at the ecosystem level.

Our study provides new insights into understanding tropical P cycling processes and modelling forest P limitation under warming. The findings of the sustained plant P demand driven by increasing P supplies through multiple biological and geochemical processes within the ecosystems (e.g., plant P resorption, soil P mineralization, and P release from Fe minerals) under warming extend the traditional assessment method (measure P supply by available/soluble P) and the current paradigm states (assess P conversation only by soil P mineralization). This suggests that previous studies may have overestimated the impacts of P limitation on ecosystem C stock under at least short-term or medium-term future warming (Wieder et al., 2015). Our study also provides additional information on P dynamics that can be incorporated into earth system models which require further quantification (Reed et al., 2015; Sun et al., 2017).

Conclusion

Our 7-year field warming study demonstrated that warming led to a net transfer of soil P from moderatelyavailable organic P and Fe-Pi to the plant without altering litter P mineralization and ecosystem P loss. Warming thus increases the plant biological "control" of the plant-soil P-cycle, increasing the stocks of P involved in the fast P plant-soil cycle and plant P-use efficiency (higher C:P ratio in litter). The results suggest that warming changed net balance of plant-P cycle through multiple processes (the enhanced plant P resorption, soil P mineralization and dissolution) rather than in single form or process, which extends our previous understanding of the climate-demand/supply hypothesis and the application of nutrient resorption theory and emphasizes the assessment of P dynamics at the ecosystem level. In this humid tropical forest under warming, greater P supply to support higher mass production is expected than previous prediction via short-term experiments or studies with single evaluation, and moderate warming may alleviate current P deficiency and/or limitation in tropical plants and contribute to greater plant C fixation in a warmer future climate.

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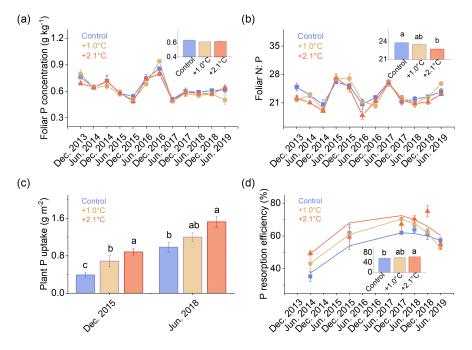


Fig. 1 Effects of warming on plant P status. Different lowercase letters indicate significant differences among treatments (P < 0.05). Error bars indicate standard errors (n = 3).

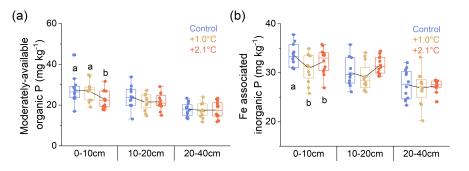


Fig. 2 Effects of warming on moderately-available P concentrations in three-layer soils (n = 12). Different lowercase letters indicate significant differences among treatments (P < 0.05).

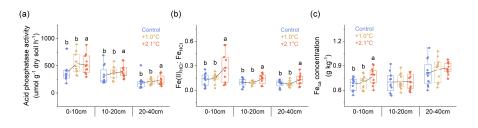


Fig. 3 Effects of warming on acid phosphatase activity (a), ratio of Fe(II) to total HCl-extractable iron (Fe_{HCl}) (b) and citrate-ascorbate extractable Fe concentrations (c) in three-layer soils (n = 12). Different lowercase letters indicate significant differences among treatments (P < 0.05).

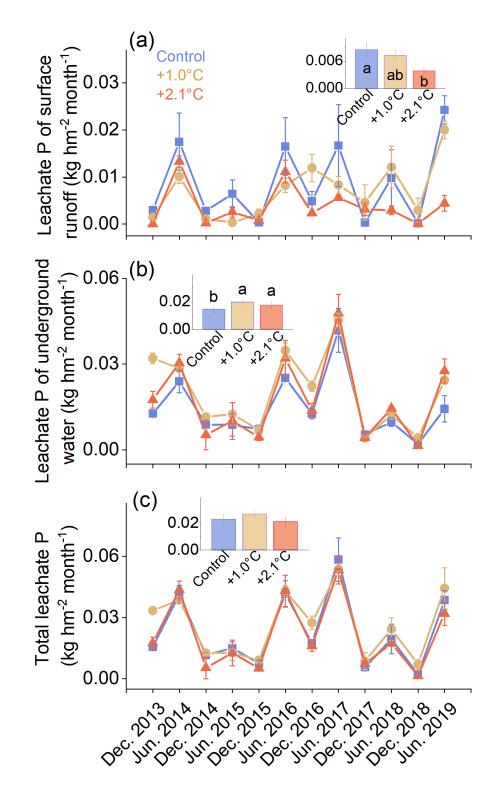


Fig. 4 Effects of warming on leachate P in surface runoff and underground water, and total P leachate (both P in runoff and groundwater). Different lowercase letters indicate significant differences among treatments (P < 0.05). Error bars indicate standard errors (n = 3).

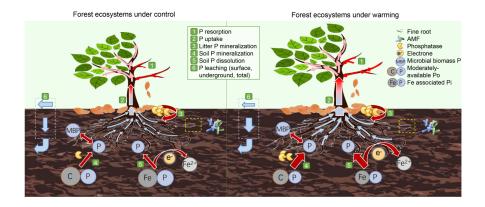


Fig. 5 Schematic showing how the P cycling links P demand and P supply during litter decomposition under control and warming. The increased plant P uptake (2) with rapid plant growth under warming is supported by the enhanced plant P resorption (1), soil P mineralization (4), and P release from Fe minerals (5) without changing litter P mineralization (3), total leachate P (6), fine root and arbuscular mycorrhizal fungi abundance (AMF) within the ecosystems. The enhanced soil P mineralization is associated with the increased phosphatase activity, while the enhanced P dissolution is indicated by the increase in reduction of Fe concomitantly released Fe associate inorganic P. The increase in released P is rapidly immobilized by plants but not microbes and results in an equilibrium of soluble P.

Table S1 Effects of warming on soil pH and bulk density in soil of 0-10 cm depth, and in-site soil redox potential at 10 cm depth. The numbers in brackets are standard errors (n = 3). Different lowercase letters indicate significant differences among treatments at P < 0.05.

		Experimental time	Experimental time	Experimental time	Experimental time	P-value	P-va
		Dec. 2017	Jun. 2018	Dec. 2018	Jun. 2019	Warming	Tim
$_{\rm pH}$	Control	4.3(0.0)	4.4(0.0)	4.1(0.1)	4.3(0.1)	0.517	0.00
	$+1.0^{\circ}\mathrm{C}$	4.3(0.1)	4.4(0.0)	4.1(0.1)	4.5(0.0)		
	$+2.1^{\circ}\mathrm{C}$	4.3(0.0)	4.4(0.0)	4.1(0.0)	4.5(0.0)		
Redox	Control	626.6(15)a	662.4(10.6)a			0.015	0.00
potential	$+1.0^{\circ}\mathrm{C}$	579.7(21.2)ab	664.4(9.1)b				
(mV)	$+2.1^{\circ}\mathrm{C}$	516.7(19.6)b	624(5.1)b				
Bulk	Control	· · · ·			1.3(0.0)	0.662	
density	$+1.0^{\circ}\mathrm{C}$				1.2(0.0)		
$(g \text{ cm}^{-3})$	$+2.1^{\circ}\mathrm{C}$				1.2(0.1)		

Table S2 Effects of warming on litter P status The numbers in brackets are standard errors (n = 3). Different lowercase letters indicate significant differences among treatments at P < 0.05.

		Experimental time	Experimental time	Experimental time	Experimental time	P
		Dec. 2017	Jun. 2018	Dec. 2018	Jun. 2019	W
Litter C:P	Control	1759(76)	1988(120)	1652(120)b	1314(55)	0
	$+1.0^{\circ}\mathrm{C}$	2427(51)	2217(120)	1834(132)b	1454(153)	
	$+2.1^{\circ}\mathrm{C}$	2176(332)	2428(269)	2564(332)a	1094(94)	
Litter P stock	Control	9.5(0.6)	10.3(0.7)	10.3(1.1)	9.7(1.5)	0.
$(g m^{-2})$	$+1.0^{\circ}\mathrm{C}$	6.9(0.6)	8.8(0.5)	9.3(0.3)	9.8(0.6)	
	$+2.1^{\circ}\mathrm{C}$	9.4(0.9)	9.4(0.7)	8.2(1.2)	11(0.4)	
Litter phosphatase	Control	11043(1008)	5660(781)	7735(2438)	6144(864)	0.

$(\text{umol g}^{-1} \text{ dry litter h}^{-1})$	$+1.0^{\circ}\mathrm{C}$	9181(1367)	7739(846)	7731(868)	6225(1011)
	$+2.1^{\circ}\mathrm{C}$	8865(1185)	5494(1801)	8200(474)	4993(301)

Table S3 Effects of warming on soil P components in three-layer soils. The values in parentheses represent standard error (n = 3). Al-Pi, Al associated inorganic P; Al-Po, Al associated organic P; Fe-Po, Fe associated organic P; MBP, microbial biomass P. Different lowercase letters indicate significant differences among treatments at P < 0.05.

	Soil	Treatment	$\label{eq:experimental} Experimental Experimental Experimental P-$					<i>P</i> -	P
	depth		time time		time	time	value	value	va
			Dec.	Jun.	Dec.	Jun.	Warming	Time	W
			2017	2018	2018	2020			
Soluble P	0-10 cm	Control	0.3(0.1)	0.1(0)	0.3(0.1)	0.4(0.1)	0.913	0.924	0.3
$(mg kg^{-1})$		$+1.0^{\circ}\mathrm{C}$	0.3(0.1)	0.2(0.1)	0.4(0.1)	0.3(0.1)			
		$+2.1^{\circ}\mathrm{C}$	0.3(0)	0.3(0.1)	0.2(0)	0.5(0)			
	10–20 cm	Control	0.2(0.1)	0.1(0)	0.1(0)	0.3(0.1)	0.394	0.461	0.0
		$+1.0^{\circ}\mathrm{C}$	0.2(0)	0.1(0)	0.3(0)	0.3(0)			
		$+2.1^{\circ}\mathrm{C}$	0.2(0)	0.1(0)	0.2(0.1)	0.5(0)			
	20–40 cm	Control	0.2(0)	0.1(0)	0.1(0)	0.2(0)	0.200	0.656	0.
		$+1.0^{\circ}\mathrm{C}$	0.3(0.1)	0.1(0)	0.2(0.1)	0.3(0.1)			
		$+2.1^{\circ}\mathrm{C}$	0.1(0)	0.1(0.1)	0.2(0.1)	0.4(0)			
Al-Pi (mg kg ⁻¹)	$0\!\!-\!\!10~\mathrm{cm}$	Control	3.5(0.3)	3.7(0.2)	4(0.6)	3.7(0.2)b	0.586	0.063	0.5
(mg kg ⁻¹)		$+1.0^{\circ}\mathrm{C}$	3.3(0.2)	4(0.3)	3(0.2)	$3.9(0)\mathrm{b}$			
0 /		$+2.1^{\circ}\mathrm{C}$	3(0.5)	4(0.4)	4.1(0.7)	4.3(0.2)a			
	10–20 cm	Control	3.3(0.3)	3(0.1)	2.6(0.2)	3.3(0.1)	0.557	0.300	0.9
	om	$+1.0^{\circ}\mathrm{C}$	3(0.1)	2.8(0.3)	3(0.2)	3.5(0.3)			
		$+2.1^{\circ}C$	3(0.2)	3.1(0.2)	3.4(0.1)	3.4(0.3)			
	20–40 cm	Control	2.6(0.5)	2.3(0.1)	1.9(0.3)	2.9(0.1)	0.469	0.080	0.4
		$+1.0^{\circ}\mathrm{C}$	2.9(0.4)	2.2(0.1)	2.3(0.3)	3(0.1)			
		$+2.1^{\circ}C$	2.1(0.2)	2.7(0.3)	2.8(0.2)	3.3(0.2)			
Al-Po	$_{ m cm}^{ m 0-10}$	Control	5.2(0.4)	2.9(1.2)	3.8(0.2)	2.8(1.1)	0.900	0.024	0.3
$(mg kg^{-1})$	ciii	$+1.0^{\circ}\mathrm{C}$	4.6(1)	3.5(0.9)	3(0.4)	4.5(0.2)			
kg)		$+2.1^{\circ}\mathrm{C}$	5(0.9)	2.4(0.5)	3.5(0.1)	3.3(0.2)			
	10–20 cm	Control	3.9(1.5)	2.8(0.4)	1.4(0.4)	3.5(0.2)	0.790	0.913	0.
	-	$+1.0^{\circ}\mathrm{C}$	3.9(0.8)	2.3(0.6)	2.1(0.9)	2.8(0.4)			
		$+2.1^{\circ}C$	3.5(0.3)	3.1(0.8)	2.9(0.7)	3.2(0.4)			
	20–40 cm	Control	2.4(0.4)	0.7(0.1)	0.9(0.3)	3(1.1)	0.903	0.098	0.3
	0111	$+1.0^{\circ}\mathrm{C}$	3.7(0.3)	0.9(0.4)	0.7(0.1)	1.5(0.6)			
		$+2.1^{\circ}C$	3.4(0.4)	0.7(0.3)	1.2(0.5)	1.3(0.3)			

Fe-Po	$_{ m cm}^{ m 0-10}$	Control	20.1(2)	21.4(0.5)	28.1(1.5)a	21.8(2.8)	0.079	0.962	0.4
$(mg kg^{-1})$	CIII	$+1.0^{\circ}\mathrm{C}$	16.4(0.1)	24.2(2.5)	27.5(2)ab	23.4(0.1)			
0 /		$+2.1^{\circ}\mathrm{C}$	18.2(3.5)	22(2.2)	19.4(2.3)b	17.6(1.5)			
	10 - 20	Control	16(2.4)	23.1(3.9)	23.8(2.5)	21.8(2.2)	0.298	0.736	0.3
	cm			~ /		~ /			
		$+1.0^{\circ}\mathrm{C}$	16.9(1.8)	19(2.4)	23(2)	15.7(2.1)			
		$+2.1^{\circ}\mathrm{C}$	16.8(2.2)	15.2(1.2)	22(2.7)	20(1.7)			
	20 - 40	Control	13(0.7)	16.5(1.6)	19.2(1.6)	15.9(1.7)	0.995	0.949	0.9
	cm								
		$+1.0^{\circ}\mathrm{C}$	13.2(1.7)	14.4(2)	18.3(1.6)	18.3(3.3)			
		$+2.1^{\circ}\mathrm{C}$	16.9(2.5)	17.2(2.8)	15.5(2.2)	14.3(2)			
MBP	0 - 10	Control	5.4(0.8)			4.4(0.7)	0.405	0.087	0.7
	cm								
$(mg kg^{-1})$		+1.0°C	4.9(1)			3.6(1.1)			
0 ,		$+2.1^{\circ}\mathrm{C}$	4.7(1.3)			3.8(1.4)			
	10 - 20	Control	3.2(2.3)			3.6(3.1)	0.080	0.569	0.7
	cm								
		$+1.0^{\circ}\mathrm{C}$	3.9(0.7)			3.4(1.7)			
		$+2.1^{\circ}\mathrm{C}$	3.1(2.4)			2.3(1.5)			
	20 - 40	Control	2.9(0.4)			3.8(1.2)	0.138	0.017	0.3
	cm								
		$+1.0^{\circ}\mathrm{C}$	4.1(1.6)			4.6(2.6)			
		$+2.1^{\circ}\mathrm{C}$	2.7(1.4)			6.8(1.4)			

Table S4 Effects of warming on soil citrate-dithionite extractable Fe in soil of 0-10 cm depth. The numbers in brackets are standard error (n = 3).

$+1.0^{\circ}\mathrm{C}$	Jun. 2014 37.5(3.6) 36.9(1.4)	Dec. 2015 33.6(1.7) 42.3(3.3)	Experimental time Dec. 2017 36.9(0.3) 34.8(2.5)	Jun. 2018 29.2 (0.5) 32 (0.8)	P-value Warming 0.433	<i>P</i> -value Time < 0.001	<i>P</i> -va War 0.15
+1.0 C +2.1°C	39(2.1)	42.3(3.5) 41.6(3.6)	36.2(2)	30.8(0.8)			

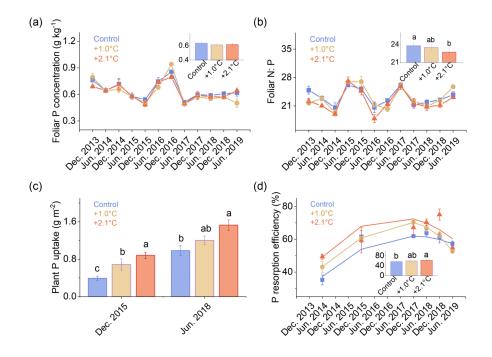


Fig. S1 The scheme of warming experimental design. In order to conduct warming experiment, we used natural temperature differences at three elevations by translocating model forest ecosystems from the high-elevation site (control, 600 m) to the two lower-elevation sites $(+1.0^{\circ}C, 300 \text{ m}; +2.1^{\circ}C, 30 \text{ m})$.

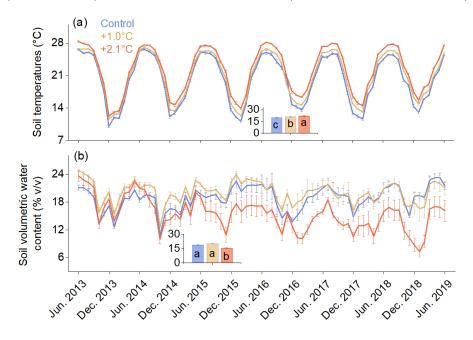


Fig. S2 Effects of warming on soil temperatures and soil moisture on monthly average at 5cm depth during 2013–2019. Different lowercase letters indicate significant differences among treatments (P < 0.05). Error bars indicate standard errors (n = 3).

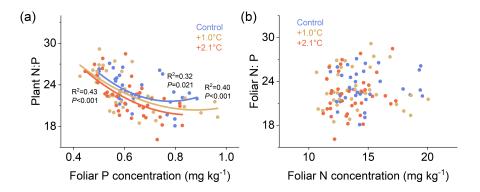


Fig. S3 Relationship between foliar N:P and foliar P and N concentration under control and warming treatments during 2013–2019.

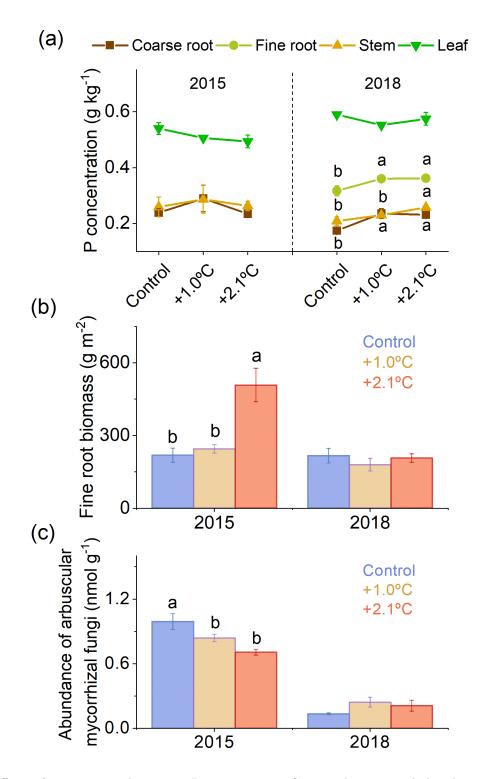


Fig. S4 Effects of warming on plant tissue P concentration, fine root biomass and abundance of arbuscular mycorrhizal fungi. Different lowercase letters indicate significant differences among treatments (P < 0.05). Error bars indicate standard errors (n = 3).

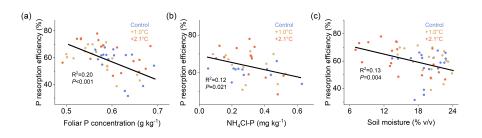


Fig. S5 Relationship between plant resorption efficiency and foliar P concentrations, NH_4Cl-P , and soil moisture under control and warming treatments during 2014–2015 and 2017–2019.